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Europe is one of the main challenges facing economics in the 1990s Combining the detached analysis of a Western specialist
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assumptions and superficial description to give a rigorous empirical analysis of the process of transition Professor Ellman
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$$\text{N}_2 + \text{F}_2 \rightarrow \text{NF}_3$$
$$\text{C}_6\text{H}_{10}\text{O}_2 + \text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{HBr} + \text{KHCO}_3 + \text{H}_2\text{O} + \text{KBr} + \text{CO}_2 + \text{GaBr} + \text{NaBr} + \text{SnO}$$
$$\text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + \text{SO}_3 \rightarrow \text{Na}_2\text{SO}_4 + \text{SnF}_2 + \text{N}_2\text{O}_3$$
using the equation from problem 2 above answer the following

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$$\text{C}_6\text{H}_{10}\text{O}_2 + 17\text{O}_2 \rightarrow 12\text{CO}_2 + 10\text{H}_2\text{O}$$
$$\text{HBr} + \text{KHCO}_3 + \text{H}_2\text{O} + \text{KBr} + \text{CO}_2 + 2\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + 3\text{SO}_3 + 6\text{NaBr} + 3\text{SnO} + 2\text{NF}_3 + 3\text{SnF}_2 + \text{N}_2\text{O}_3$$
using the following equation
$$2\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$$

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$$\text{Ce} + \text{Zn} + \text{CuCl}_2 \rightarrow \text{ZnCl}_2 + \text{Cu} + \text{Zn} + \text{CuCl}_x$$
2 ZnCl_x 2 Cu how many moles of Ce ZnCl₂ ZnCl_x 2 will be produced from 23.0 g of Ce Zn Zn assuming Ce CuCl₂ CuCl_x 2 is available in excess

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$$\text{Br}_2 + \text{N}_2 + 3\text{F}_2 + 2\text{NF}_3 + \text{Br}_2 + \text{C}_6\text{H}_{10}\text{O}_2 + 17\text{O}_2 + 12\text{CO}_2 + 10\text{H}_2\text{O} + \text{Br}_2 + \text{HBr} + \text{KHCO}_3 + \text{H}_2\text{O} + \text{KBr} + \text{CO}_2 + \text{Br}_2 + 2\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + 3\text{SO}_3 + 6\text{NaBr} + \text{Br}_2 + 5\text{SO}_2 + 2\text{NF}_3 + 3\text{SnF}_2 + \text{N}_2\text{O}_3 + \text{Br}_2$$
using the following equation
$$\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$$

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