

Periodic Table of Elements (Worksheet)

Periodic Table of Elements Worksheet

Q1. Using the following activity table, sort the following elements into groups:
1. Helium 2. Hydrogen 3. Oxygen
4. Chlorine 5. Nitrogen 6. Carbon
7. Bromine 8. Fluorine 9. Phosphorus

Group 1	Group 2	Group 3	Group 4
He, H, O, N, P	C, Br, F		

Q2. Write the name of each element:
1. The most abundant element in the universe is _____
2. The second most abundant element in the universe is _____
3. The third most abundant element in the universe is _____
4. The fourth most abundant element in the universe is _____
5. The fifth most abundant element in the universe is _____
6. The sixth most abundant element in the universe is _____
7. The seventh most abundant element in the universe is _____
8. The eighth most abundant element in the universe is _____
9. The ninth most abundant element in the universe is _____
10. The tenth most abundant element in the universe is _____

Q3. What is the atomic number of Helium?
A. 1 B. 2 C. 3 D. 4

Q4. What is the atomic number of Hydrogen?
A. 1 B. 2 C. 3 D. 4

Q5. What is the atomic number of Oxygen?
A. 1 B. 2 C. 3 D. 4

Q6. What is the atomic number of Chlorine?
A. 1 B. 2 C. 3 D. 4

Q7. What is the atomic number of Nitrogen?
A. 1 B. 2 C. 3 D. 4

Q8. What is the atomic number of Carbon?
A. 1 B. 2 C. 3 D. 4

Q9. What is the atomic number of Bromine?
A. 1 B. 2 C. 3 D. 4

Q10. What is the atomic number of Fluorine?
A. 1 B. 2 C. 3 D. 4

Periodic Table of Elements Worksheet

Q1. Name the following elements:
1. Helium 2. Hydrogen 3. Oxygen
4. Chlorine 5. Nitrogen 6. Carbon
7. Bromine 8. Fluorine 9. Phosphorus

Q2. Complete the following sentence using the words below:
1. Element 2. Atom 3. Molecule
4. Compound 5. Ion 6. Isotope

Q3. Complete the following table:

Element	Symbol	Atomic Number
Hydrogen	H	1
Helium	He	2
Oxygen	O	8
Chlorine	Cl	17
Nitrogen	N	7
Carbon	C	6
Bromine	Br	35
Fluorine	F	9
Phosphorus	P	15

Q4. Complete the following sentence using the words below:
1. Element 2. Atom 3. Molecule
4. Compound 5. Ion 6. Isotope

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