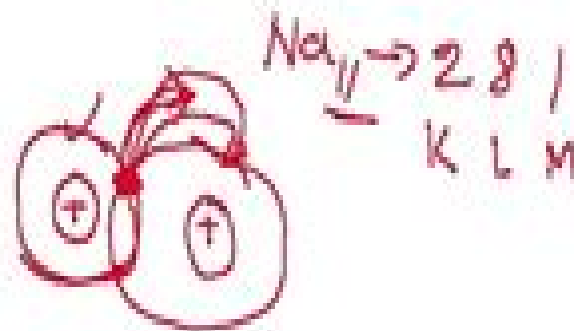
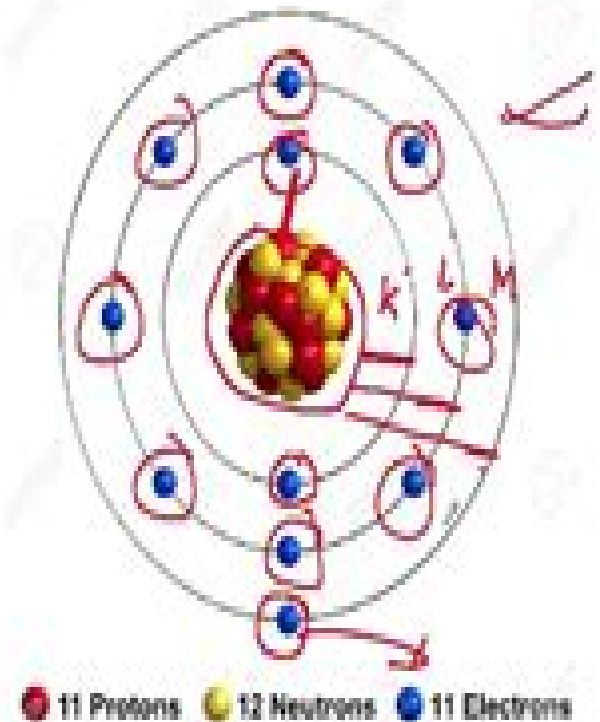
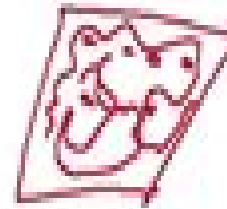


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1. Metal possesses free electrons.
2. In solids atoms are closely packed hence the orbitals are overlap each other, so the valence electron do not localized. It jump from one orbital to the another orbital of the other atom and continuously move on the surface of the metals with certain energy.
3. It is also called conduction electron.
4. The movement of electron in the metal surface does not loss any energy
5. The free electrons move randomly on the surface of metals like gases atom hence it is called free electron gas



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