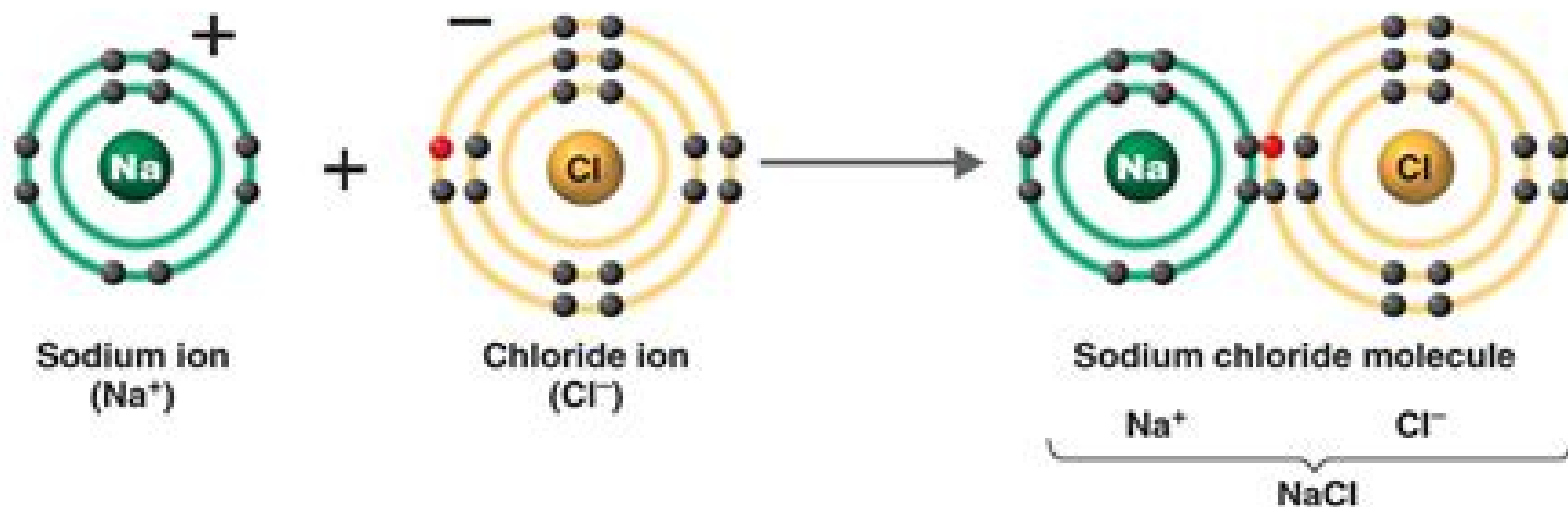


**(a)** A sodium atom (Na) loses one electron to an electron acceptor and forms a sodium ion ( $\text{Na}^+$ ). A chlorine atom (Cl) accepts one electron from an electron donor to become a chloride ion ( $\text{Cl}^-$ ).



**(b)** The sodium and chloride ions are attracted because of their opposite charges and are held together by an ionic bond to form a molecule of sodium chloride.

# Electronic Structure Chemical Bonding

**Minjie Lin**



## **Electronic Structure Chemical Bonding:**

## Whispering the Techniques of Language: An Emotional Journey through **Electronic Structure Chemical Bonding**

In a digitally-driven world wherever displays reign supreme and immediate communication drowns out the subtleties of language, the profound techniques and emotional nuances hidden within words often go unheard. Yet, situated within the pages of **Electronic Structure Chemical Bonding** a interesting literary value blinking with natural thoughts, lies a fantastic journey waiting to be undertaken. Written by a talented wordsmith, this wonderful opus attracts readers on an introspective journey, softly unraveling the veiled truths and profound affect resonating within the very fabric of each word. Within the mental depths of the touching review, we can embark upon a heartfelt exploration of the book is core themes, dissect its captivating writing model, and fail to the effective resonance it evokes strong within the recesses of readers hearts.

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