

THE PERIODIC TABLE OF ELEMENTS

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1	<div><div>1</div><div>H</div><div>Hydrogen</div></div>																	<div><div>2</div><div>He</div><div>Helium</div></div>	
2	<div><div>3</div><div>Li</div><div>Lithium</div></div>	<div><div>4</div><div>Be</div><div>Beryllium</div></div>																	<div><div>10</div><div>Ne</div><div>Neon</div></div>
3	<div><div>11</div><div>Na</div><div>Sodium</div></div>	<div><div>12</div><div>Mg</div><div>Magnesium</div></div>																	<div><div>18</div><div>Ar</div><div>Argon</div></div>
4	<div><div>19</div><div>K</div><div>Potassium</div></div>	<div><div>20</div><div>Ca</div><div>Calcium</div></div>	<div><div>21</div><div>Sc</div><div>Scandium</div></div>	<div><div>22</div><div>Ti</div><div>Titanium</div></div>	<div><div>23</div><div>V</div><div>Vanadium</div></div>	<div><div>24</div><div>Cr</div><div>Chromium</div></div>	<div><div>25</div><div>Mn</div><div>Manganese</div></div>	<div><div>26</div><div>Fe</div><div>Iron</div></div>	<div><div>27</div><div>Co</div><div>Cobalt</div></div>	<div><div>28</div><div>Ni</div><div>Nickel</div></div>	<div><div>29</div><div>Cu</div><div>Copper</div></div>	<div><div>30</div><div>Zn</div><div>Zinc</div></div>	<div><div>31</div><div>Ga</div><div>Gallium</div></div>	<div><div>32</div><div>Ge</div><div>Germanium</div></div>	<div><div>33</div><div>As</div><div>Arsenic</div></div>	<div><div>34</div><div>Se</div><div>Selenium</div></div>	<div><div>35</div><div>Br</div><div>Bromine</div></div>	<div><div>36</div><div>Kr</div><div>Krypton</div></div>	
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Atomic Number → 1
Symbol → H
Name → Hydrogen
Average Atomic Mass → 1.008

- Alkali Metals
- Alkaline Earth Metals
- Transition Metals
- Other Metals
- Metalloids
- Non-metals
- Halogens
- Noble Gases
- Lanthanides
- Actinides

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