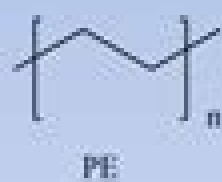


Thermal Degradation

Depending upon the presence of oxygen, temperature and structure of polymer, degradation and/or oxidation reactions will occur.

Theoretical point of view most commercial polymer systems should be relatively stable above their melting point in the absence of oxygen.

It is interesting to note that saturated hydrocarbons are much more stable than polyethylene (PE) in the absence of oxygen as are chloroalkanes when compared with PVC.



$$T_{\text{deg PE}} < T_{\text{deg Butadecane}}$$

or



$$T_{\text{deg PVC}} < T_{\text{deg 2,4,6-trichloroheptane}}$$

In some cases this temperature difference may be as high as 200°C. There are mainly two reasons for this difference in behavior;

- The first of which is simply that polymers by virtue of long chain nature are able to breakdown into smaller molecular fragments i.e. monomer formation via unzipping reactions

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