



Elementary Quantum Mechanics

Hilbrand Johannes Groenewold



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Elementary Quantum Mechanics David S. Saxon, 2012-01-01 Based on lectures for an undergraduate UCLA course in quantum mechanics this volume focuses on the formulas of quantum mechanics rather than applications Widely used in both upper level undergraduate and graduate courses it offers a broad self contained survey rather than in depth treatments Topics include the dual nature of matter and radiation state functions and their interpretation linear momentum the motion of a free particle Schrödinger's equation approximation methods angular momentum and many other subjects In the interests of keeping the mathematics as simple as possible most of the book is confined to considerations of one dimensional systems A selection of 150 problems many of which require prolonged study amplify the text's teachings and an appendix contains solutions to 50 representative problems This edition also includes a new Introduction by Joseph A Rudnick and Robert Finkelstein Reprint of the Holden Day Inc San Francisco 1968 edition **Elementary Quantum Mechanics** ,1962

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On the Principles of Elementary Quantum Mechanics Hilbrand Johannes Groenewold, 2014-01-15 **The Development of Elementary Quantum Theory** Herbert Capellmann, 2017-09-12 This book traces the evolution of the ideas that eventually resulted in the elementary quantum theory in 1925-26. Further, it discusses the essential differences between the fundamental equations of Quantum Theory derived by Born and Jordan, logically comprising Quantum Mechanics and Quantum Optics, and the traditional view of the development of Quantum Mechanics. Drawing on original publications and letters written by the main protagonists of that time, it shows that Einstein's contributions from 1905 to 1924 laid the essential foundations for the development of Quantum Theory. Einstein introduced quantization of the radiation field. Born added quantized mechanical behavior. In addition, Born recognized that Quantum Mechanics necessarily required Quantum Optics; his radical concept of truly discontinuous and statistical quantum transitions (quantum leaps) was directly based on Einstein's physical concepts.

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elementary level except possibly for tensor products which are sadly neglected many linear algebra courses today but which are literally exploding in a variety of current applications Elementary Quantum Mechanics in One Dimension Robert Gilmore, 2004-10-22 One of the key components of modern physics quantum mechanics is used in such fields as chemistry electrical engineering and computer science Central to quantum mechanics is Schrödinger's Equation which explains the behavior of atomic particles and the energy levels of a quantum system Robert Gilmore's innovative approach to Schrödinger's Equation offers new insight into quantum mechanics at an elementary level Gilmore presents compact transfer matrix methods for solving quantum problems that can easily be implemented on a personal computer He shows how to use these methods on a large variety of potentials both simple and periodic He shows how to compute bound states scattering states and energy bands and describes the relation between bound and scattering states Chapters on alloys superlattices quantum engineering and solar cells indicate the practical application of the methods discussed Gilmore's concise and elegant treatment will be of interest to students and professors of introductory and intermediate quantum courses as well as professionals working in electrical engineering and applied mathematics *Elementary Quantum Chemistry, Second Edition* Frank L. Pilar, 2013-08-22 As the author notes in the Preface to this valuable text experimental chemists have moved past studying the average behavior of atoms or molecules to probe the step by step behavior of individual atoms and molecules as they collide form transition states and ultimately form products In such experiments quantum mechanical computations do two useful tasks They fill in the observational gaps and help to interpret what has been observed This introductory course developed by the former chairman of the chemistry department at the University of New Hampshire covers among other topics the origins of the quantum theory the Schrödinger wave equation the quantum mechanics of simple systems the rigid rotator the hydrogen atom electron spin and many electron systems the quantum states of atoms the Hartree Fock self consistent field method the electronic structure of molecules and semi empirical molecular orbital methods One of the great values of the course is its calculations and diagrams which were created specifically for this text and which students will be able to replicate on their home computers The text will be most useful for advanced undergraduate or beginning graduate students who have had calculus and at least a year of undergraduate physics A knowledge of differential equations linear algebra and atomic physics is helpful but not essential Seven appendices give a concise exposition of mathematical and physical material that may not be part of the students background *Elementary Quantum Mechanics* Morten Scharff, 1969 *Elementary Quantum Mechanics* William Judson LeVeque, 1962 **Elementary Quantum Chemistry** Frank L. Pilar, 1990 Useful introductory course covers origins of quantum theory Schrödinger wave equation quantum mechanics of simple systems electron spin the quantum states of atoms the Hartree Fock self consistent field method more

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