

Periodic Chart of the Elements

IA																		NOBLE GASES	
1	<div>1</div> <div>H</div> <div>1.008</div>																	<div>2</div> <div>He</div> <div>4.003</div>	
2	<div>3</div> <div>Li</div> <div>6.941</div>	<div>4</div> <div>Be</div> <div>9.012</div>																	
3	<div>11</div> <div>Na</div> <div>22.990</div>	<div>12</div> <div>Mg</div> <div>24.305</div>																	
4	<div>19</div> <div>K</div> <div>39.098</div>	<div>20</div> <div>Ca</div> <div>40.078</div>	<div>21</div> <div>Sc</div> <div>44.956</div>	<div>22</div> <div>Ti</div> <div>47.88</div>	<div>23</div> <div>V</div> <div>50.942</div>	<div>24</div> <div>Cr</div> <div>52.00</div>	<div>25</div> <div>Mn</div> <div>54.938</div>	<div>26</div> <div>Fe</div> <div>55.845</div>	<div>27</div> <div>Co</div> <div>58.933</div>	<div>28</div> <div>Ni</div> <div>58.69</div>	<div>29</div> <div>Cu</div> <div>63.546</div>	<div>30</div> <div>Zn</div> <div>65.38</div>	<div>31</div> <div>Ga</div> <div>69.723</div>	<div>32</div> <div>Ge</div> <div>72.64</div>	<div>33</div> <div>As</div> <div>74.922</div>	<div>34</div> <div>Se</div> <div>78.96</div>	<div>35</div> <div>Br</div> <div>79.904</div>	<div>36</div> <div>Kr</div> <div>83.80</div>	
5	<div>37</div> <div>Rb</div> <div>85.468</div>	<div>38</div> <div>Sr</div> <div>87.62</div>	<div>39</div> <div>Y</div> <div>88.906</div>	<div>40</div> <div>Zr</div> <div>91.224</div>	<div>41</div> <div>Nb</div> <div>92.906</div>	<div>42</div> <div>Mo</div> <div>95.94</div>	<div>43</div> <div>Tc</div> <div>98.906</div>	<div>44</div> <div>Ru</div> <div>101.07</div>	<div>45</div> <div>Rh</div> <div>102.91</div>	<div>46</div> <div>Pd</div> <div>106.42</div>	<div>47</div> <div>Ag</div> <div>107.868</div>	<div>48</div> <div>Cd</div> <div>112.411</div>	<div>49</div> <div>In</div> <div>114.818</div>	<div>50</div> <div>Sn</div> <div>118.710</div>	<div>51</div> <div>Sb</div> <div>121.757</div>	<div>52</div> <div>Te</div> <div>127.6</div>	<div>53</div> <div>I</div> <div>126.905</div>	<div>54</div> <div>Xe</div> <div>131.29</div>	
6	<div>55</div> <div>Cs</div> <div>132.905</div>	<div>56</div> <div>Ba</div> <div>137.327</div>	<div>57</div> <div>La</div> <div>138.905</div>	<div>72</div> <div>Hf</div> <div>178.49</div>	<div>73</div> <div>Ta</div> <div>180.948</div>	<div>74</div> <div>W</div> <div>183.84</div>	<div>75</div> <div>Re</div> <div>186.207</div>	<div>76</div> <div>Os</div> <div>190.23</div>	<div>77</div> <div>Ir</div> <div>192.222</div>	<div>78</div> <div>Pt</div> <div>195.084</div>	<div>79</div> <div>Au</div> <div>196.967</div>	<div>80</div> <div>Hg</div> <div>200.59</div>	<div>81</div> <div>Tl</div> <div>204.383</div>	<div>82</div> <div>Pb</div> <div>207.2</div>	<div>83</div> <div>Bi</div> <div>208.980</div>	<div>84</div> <div>Po</div> <div>209</div>	<div>85</div> <div>At</div> <div>210</div>	<div>86</div> <div>Rn</div> <div>222</div>	
7	<div>87</div> <div>Fr</div> <div>[223]</div>	<div>88</div> <div>Ra</div> <div>[226]</div>	<div>89</div> <div>Ac</div> <div>[227]</div>	<div>104</div> <div>Unq</div> <div>[261]</div>	<div>105</div> <div>Unp</div> <div>[262]</div>	<div>106</div> <div>Unh</div> <div>[263]</div>	<div>107</div> <div>Uns</div> <div>[264]</div>												

THE SYMBOL. Shown in the middle of each block directly below the name of the element.
The color used indicates the physical state of the element under ordinary conditions: black for solids, green for liquids and blue for gases.

THE ATOMIC WEIGHT. Shown below the symbol for each element - the atomic weight is shown in black. The values are taken from the official Report on Atomic Weights (J. J. Auer, Chem. Soc. 1913-1914). For elements not listed in the Report the mass number of the longest lived isotopes is shown in brackets.

THE ATOMIC NUMBER. Shown in red in the upper left hand corner.

ELECTRONIC CONFIGURATION. Shown at the upper right as a group of black numbers. When read downward they indicate the number of electrons normally found in successive energy levels.

58	59	60	61	62	63	64	65	66	67	68	69	70	71
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu

90	91	92	93	94	95	96	97	98	99	100	101	102	103
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

Elements Of Old English

Ying Liu



Elements Of Old English:

Unveiling the Power of Verbal Art: An Psychological Sojourn through **Elements Of Old English**

In some sort of inundated with displays and the cacophony of fast conversation, the profound power and emotional resonance of verbal art frequently diminish into obscurity, eclipsed by the regular assault of sound and distractions. However, situated within the lyrical pages of **Elements Of Old English**, a captivating work of fictional brilliance that pulses with fresh thoughts, lies an memorable journey waiting to be embarked upon. Written by way of a virtuoso wordsmith, that exciting opus courses readers on a mental odyssey, lightly exposing the latent possible and profound affect embedded within the elaborate internet of language. Within the heart-wrenching expanse of the evocative analysis, we will embark upon an introspective exploration of the book is main themes, dissect its charming publishing design, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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