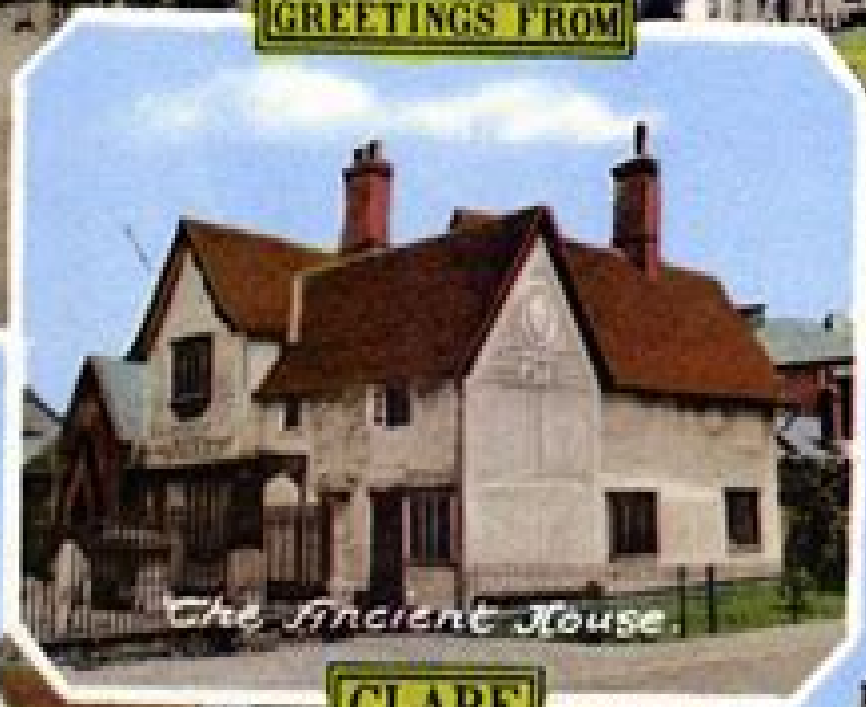




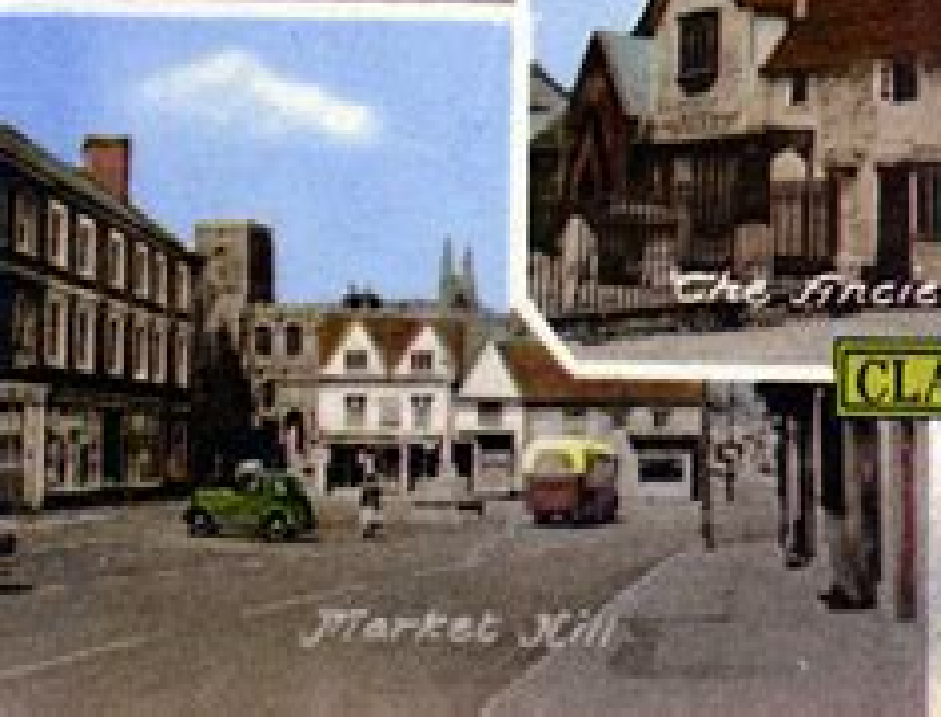
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Words selected ... by the Revd G. H. Drummond. The Music Selected Adapted&Composed by E. Miller. (Addenda, containing some additional favorite Tunes.). Edward Miller,1790 Memorials of Cambridge. (With photographs printed by F. Frith.) A new edition (entirely re-written), by C. H. Cooper Thomas Wright,1860 The Europa World of Learning ,2005 **The British Friend** ,1861 Once a Man - Twice a Boy Claude Banks,2014-06-28 Open this book and read whatever chapter appears and you will be transported back in time Tardis style into village life around Pertenhall and Kimbolton over the last 80 years You will immediately become part of the scene Claude Banks treats readers to a rich potpourri of stories of village life with a wicked eye for detail Feel the austerity of the agricultural depression leading up to World War 2 Giggle at the exploits of the local cricketers and footballers we imagine we are in Claude s classroom at Kimbolton Grammar School in the late 1930s Stories are told with understated humour including his recollections of service in the Home Guard under the command of a real life Captain Mainwaring Indeed all seemed to change with the advent of World War 2 especially with the influx of American Air Force personnel and how they brightened up local village life soon to be followed by Italian and German prisoners of war who worked on the land There are many often inadvertent comparisons to life as it is now Whilst appreciative of past times Claude is not wedded to them and shows huge adaptability to the challenges that every decade has put before him He is a true old fashioned naturalist and treats readers to stories about nature and country pursuits especially shooting and horse racing Overall the book provides superb insights into past country life with its full range of joy and tears and a cast of characters that Chaucer would have been proud of *The Genealogist's Internet* Peter Christian,2005-08-30 There is a wealth of data on the Web for those researching a family tree However the sheer volume of information and diversity of websites can make starting your search confusing and time consuming Help is at hand with this clear and authoritative guide from the National Archives The Genealogist s Internet starts by explaining how the internet works for the beginner It continues by detailing the major sources of primary data available to family historians on line and highlights the most helpful directories and gateways Use it to contact others with the same surname or to access the numerous forums discussion groups mailing lists and newsgroups focusing on genealogy

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction eas particularly nitration of methyl benzoate go give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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then actual is your actual findings yield actual theoretical x 100 what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 o

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an

optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

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lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with wither a

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