

## General ❖ Answers ❖

- G1.** E 1 rad = 100 ergs / gm; 1 Gy = 100 rads.
- G2.** E
- G3.** A
- G4.** E
- G5.** D
- G6.** B The nucleus of a hydrogen atom is a proton.
- G7.** A During positron emission, a positron and an anti-neutrino are emitted.
- G8.** E Electrons fired at a high  $Z$  target create bremsstrahlung and characteristic x-rays.
- G9.** A When a positron and an electron combine in an annihilation interaction, their rest masses (0.51 MeV each) are converted into two 0.51 MeV photons emitted in opposite directions. This is the basis of the PET scanner.
- G10.** D Tritium (H-3) has one proton and two neutrons in the nucleus.
- G11.** D Neutrons are not charged particles and, therefore, cannot ionize atoms directly. However, they transfer some of their energy to protons or light nuclei, which then cause ionization. Thus, they are indirectly ionizing.
- G12.** C Most elements can have several stable and radioactive forms. An element is characterized by the number of protons ( $Z$ ), which equals the number of electrons; the neutron number, however, can vary and is generally greater than  $Z$ .
- G13.** C  $N$  increases, but  $Z$  remains the same, so the product is an isotope. An example is  $\text{Co-59} \rightarrow \text{Co-60}$ . An isotone has the same  $N$  (neutron) number, and an isobar has the same mass number  $A$  ( $=N + Z$ ).
- G14.** D The mass number  $A$  represents the number of nucleons, i.e., protons plus neutrons.

# Finance November 2002 Exam Question And Answers

**Judith Summer**



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