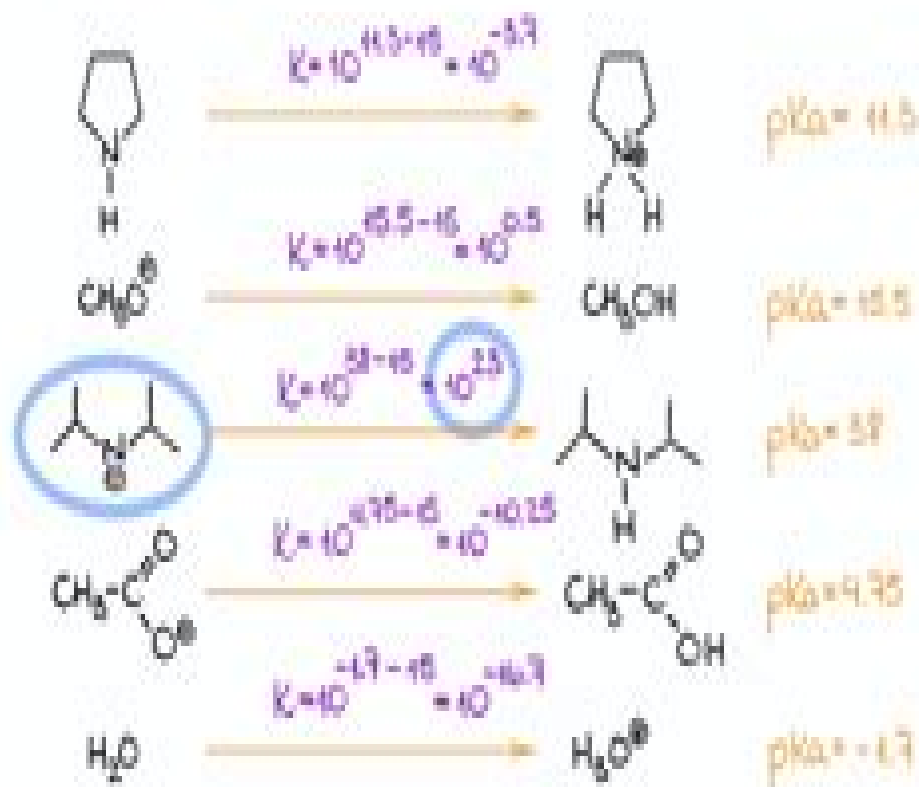


To solve this question, we need to consider the following equilibrium:



1. Set up the stem for your acid-base equilibrium

2. Write the conjugate acids for your bases

3. Find the pKa values of those conjugate acids

4. Calculate the K values by using

$$K = 10^{pKa(C.B.) - pKa(A)} \quad \text{equation}$$

5. Compare your K values to the K value in the question

Equilibrium Freshman Chemistry Problems And How To Solve Them Part 2

R Barnett



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