

diffusion equation, elementary

The equation obtained by using Fick's law to relate the current density to the neutron density in a continuity equation.

diffusion length

A length appearing in the description of diffusion when Fick's law applies. It is the square of the diffusion length (called the *diffusion area*) equal to one-sixth the mean square distance traveled by a particle of a given type or class from appearance to disappearance within an infinite homogeneous medium.

diffusion, molecular (isotopes separation)

A process whereby one substance gradually interpenetrates another substance as a result of the continuous thermal motion of the individual molecules. Diffusion due to molecular motion satisfies a partial differential equation of the form

$$\frac{\partial \rho}{\partial t} = D \frac{\partial^2 \rho}{\partial x^2}$$

where ρ is the density of the material diffusion, t is the time and x is distance in the direction of the density gradient. The quantity D is known as the **coefficient of diffusion** for the mixture in question; it turns out to be rate-determining in all separation processes using gases or liquids at pressures where molecular collisions are a significant phenomenon.

diffusivity (chemical engineering)

The relative rate of flow per unit area of a particular constituent of a mixture divided by the gradient of composition, temperature or other property considered to be causing the diffusion. Also *diffusion coefficient*.

diploid (biophysics)

Having two complete sets of chromosomes, the normal somatic number of chromosomes in higher organisms, as opposed to the half number haploid, (q.v.) found in mature germ cells.

Dirac δ

(See δ)

direct radiation (radiology)

All radiation other than the useful beam emanating from the x-ray tube anode or the inside of the tube or tube container. (See *useful beam*)

disadvantage factor

The ratio of the average neutron flux in a cell to the average neutron flux within the fissionable material.

discomposition

The process in which an atom is knocked out of its position in a crystal lattice by direct nuclear impact, e.g., by fast neutrons or by fast ions which have been previously knocked out of their lattice positions. The atom so displaced eventually comes to rest at an interstitial position or at a lattice edge.

discomposition effect

Change in chemical or physical properties resultant from discomposition. Also called *Wigner effect*.

discriminator

(See *pulse height discriminator*)

disintegration

A spontaneous nuclear transformation (radioactivity) characterized by the emission of energy from the nucleus. When numbers of nuclei are involved the process is characterized by a definite half-life. (See *radioactive half-life*) (Chemistry)

1. Radioactive decay.

2. The transformation of one nuclide into one or more different nuclides

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