

# GENERAL CHEMISTRY 1 LAB REPORT

Experiment #: 2

Title: Chemical Reactions

Purpose: Observation of different types of chemical reactions and their results.

Procedure: Reactions performing.

Chemical Reactions : 1) Double displacement reaction

2) Acid-base double displacement reaction 3) Single displacement reaction 4) Combustion 5) Composition 6) Decomposition

Reversible reactions

When chemical reaction takes place one of the results can occur :

- 1) Heat is given off
- 2) Precipitate is formed
- 3) Gas is given off
- 4) Water can be formed as a product
- 5) Change in color can take place.

Data/Results/Calculations:

1) Double displacement reaction :

This is the type of reaction in which positive and negative ions gets exchanged or interchanged,

Examples:  $\text{HCl} + \text{AgNO}_3 = \text{AgCl} + \text{HNO}_3$

here, both solutions were liquid and formed a solid precipitate. And as we have mentioned earlier that during this reaction this type of change can occur.

$\text{KI} + \text{Pb}(\text{NO}_3)_2$

This is a Double displacement reaction

here, in change will occur it will change to yellow and solid will be formed as a product. Here,

# General Chem 7e Lab

**William Jay Hale**



## General Chem 7e Lab:

*Exploring General Chemistry in the Laboratory* Colleen F. Craig, Kim N. Gunnerson, 2017-02-01 This laboratory manual is intended for a two semester general chemistry course The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life This lab manual covers topics such as composition of compounds reactivity stoichiometry limiting reactants gas laws calorimetry periodic trends molecular structure spectroscopy kinetics equilibria thermodynamics electrochemistry intermolecular forces solutions and coordination complexes By the end of this course you should have a solid understanding of the basic concepts of chemistry which will give you confidence as you embark on your career in science *World Directory of Crystallographers* Yves Epelboin, 2013-11-11

The 9th edition of the *World Directory of Crystallographers and of Other Scientists Employing Crystallographic Methods* which contains 7907 entries embracing 72 countries differs considerably from the 8th edition published in 1990 The content has been updated and the methods used to acquire the information presented and to produce this new edition of the Directory have involved the latest advances in technology The Directory is now also available as a regularly updated electronic database accessible via e mail Telnet Gopher World Wide Web and Mosaic Full details are given in an Appendix to the printed edition **Announcements for Register of Alumni** University of Michigan. College of Pharmacy, 1901

**Columbia University Bulletin** Columbia University, 1923 **American Universities and Colleges**, 2014-10-08 No detailed description available for American Universities and Colleges **Misc**, 1954 Research in Education, 1968-07

**Bulletin ... Misc[ellaneous]** United States. Office of Education, 1958 Public School Finance Programs of the United States, 1957-58 Albert Ralph Munse, Eugene Peter McLoone, 1960 *Catalogue* Howard University, 1918 *General Catalogue* Marquette University, 1919 Catalogue of the University of Alabama ... and Announcements University of Alabama, 1929 **Bulletin** University of Wisconsin, 1929 *The University of Virginia Record* University of Virginia, 1920

**Catalogue** Ohio State University, 1897 Teaching and Learning in the School Chemistry Laboratory Avi Hofstein, Muhamad Hugerat, 2021-11-05 Research into the educational effectiveness of chemistry practical work has shown that the laboratory offers a unique mode of instruction assessment and evaluation Laboratory work is an integral and important part of the learning process used to encourage the development of high order thinking and learning alongside high order learning and thinking skills such as argumentation and metacognition Authored by renowned experts in the field of chemistry education this book provides a holistic approach to cover all issues related to learning and teaching in the chemistry laboratory With sections focused on developing the skill sets of teachers as well as approaches to supporting students in the laboratory the book offers a comprehensive look at vicarious instruction methods teacher and students roles and the blend with ICT simulations and other effective approaches to practical work The book concludes with a focus on retrospective issues followed up with a look to the future of laboratory learning A product of nearly fifty years of research

this book will be useful for chemistry teachers curriculum developers researchers in chemistry education and professional development providers      **Boston University Bulletin** Boston University,1929      *Catalogue of Ursinus College, Collegeville, Montgomery County, Pennsylvania for the Academic Year ...* Ursinus College,1927      *A Laboratory Manual of General Chemistry* William Jay Hale,1917      **Guidelines for Laboratory Design** Louis J. DiBerardinis,Janet S. Baum,Melvin W. First,Gari T. Gatwood,Anand K. Seth,2013-04-08 Proven and tested guidelines for designing ideal labs for scientific investigations Now in its Fourth Edition *Guidelines for Laboratory Design* continues to enable readers to design labs that make it possible to conduct scientific investigations in a safe and healthy environment The book brings together all the professionals who are critical to a successful lab design discussing the roles of architects engineers health and safety professionals and laboratory researchers It provides the design team with the information needed to ask the right questions and then determine the best design while complying with current regulations and best practices *Guidelines for Laboratory Design* features concise straightforward advice organized in an easy to use format that facilitates the design of safe efficient laboratories Divided into five sections the book records some of the most important discoveries and achievements in Part IA Common Elements of Laboratory Design sets forth technical specifications that apply to most laboratory buildings and modules Part IB Common Elements of Renovations offers general design principles for the renovation and modernization of existing labs Part II Design Guidelines for a Number of Commonly Used Laboratories explains specifications best practices and guidelines for nineteen types of laboratories with three new chapters covering nanotechnology engineering and autopsy labs Part III Laboratory Support Services addresses design issues for imaging facilities support shops hazardous waste facilities and laboratory storerooms Part IV HVAC Systems explains how to heat cool and ventilate labs with an eye towards energy conservation Part V Administrative Procedures deals with bidding procedures final acceptance inspections and sustainability The final part of the book features five appendices filled with commonly needed data and reference materials This Fourth Edition is indispensable for all laboratory design teams whether constructing a new laboratory or renovating an old facility to meet new objectives

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