

## GENERAL INORGANIC CHEMISTRY

### Study of Matter

#### A. Classification of Matter

##### 1. Pure Substance

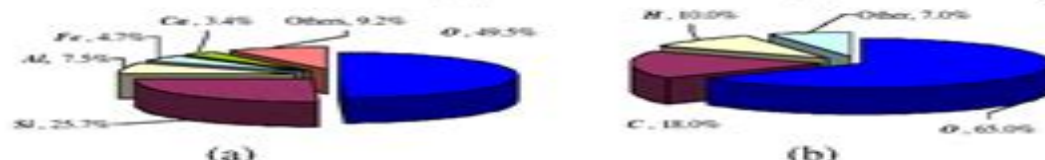
- form of matter that has distinct composition and properties

##### a. Elements

- Substance composed of one kind of atom and cannot be decomposed into simpler substances
- There are 118 known elements of which 91 elements exist in the Earth in macroscopic or trace quantities, 3 elements (Tc, Pm and Np) exist as a result of rare decay of heavy elements and 24 elements (Am to Uuo) are synthetically prepared

##### b. Compounds

- Substance formed when elements interact with other elements
- A pure compound is made up of the elements in the same proportion by mass – *Law of Definite Proportions* by Joseph Louis Proust (1754-1826)
- Abundance of the elements in (a) earth's crust and (b) human body



##### 2. Mixture

- form of matter that consists of different substances and whose composition varies
- a. **Heterogeneous mixture** – mixture that does not have uniform composition, properties and appearance
- b. **Homogeneous mixture** – mixture that is uniform throughout

#### B. Properties of Matter

1. **Physical properties** – observed without changing the identity and composition of the substance
2. **Chemical properties** – observed when a substance react to form other substances
3. **Intensive properties** – properties independent of the amount of substance present such as melting point, density, temperature, etc.
4. **Extensive properties** – properties dependent of the amount of substance present such as mass and volume

#### C. Changes in Matter

1. **Physical change** – changes that does not result to formation of new substances
2. **Chemical change** – changes that result to formation of new substances

### Atomic Theory

#### A. Dalton's Atomic Theory

- Proposed by John Dalton (1766-1844)

# General Inorganic Chemistry

**Richard Bailey**



## **General Inorganic Chemistry:**

This book delves into General Inorganic Chemistry. General Inorganic Chemistry is a vital topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into General Inorganic Chemistry, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to General Inorganic Chemistry
- Chapter 2: Essential Elements of General Inorganic Chemistry
- Chapter 3: General Inorganic Chemistry in Everyday Life
- Chapter 4: General Inorganic Chemistry in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of General Inorganic Chemistry. This chapter will explore what General Inorganic Chemistry is, why General Inorganic Chemistry is vital, and how to effectively learn about General Inorganic Chemistry.
3. In chapter 2, the author will delve into the foundational concepts of General Inorganic Chemistry. This chapter will elucidate the essential principles that must be understood to grasp General Inorganic Chemistry in its entirety.
4. In chapter 3, the author will examine the practical applications of General Inorganic Chemistry in daily life. The third chapter will showcase real-world examples of how General Inorganic Chemistry can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of General Inorganic Chemistry in specific contexts. The fourth chapter will explore how General Inorganic Chemistry is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about General Inorganic Chemistry. The final chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of General Inorganic Chemistry.

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