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# General Lebed Zagadka Robii

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these are the two types of bonds out of which every compound is made of 2 example of compounds example of compounds  
includes water  $\text{H}_2\text{O}$  hydrogen peroxide  $\text{H}_2\text{O}_2$  etc

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water are composed of molecules that is they are molecular compounds sodium chloride on the other hand contains ions it is  
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molecular compounds covalent compounds result when atoms share rather than transfer gain or lose electrons covalent  
bonding is an important and extensive concept in chemistry and it will be treated in considerable

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it is ionic write the symbols for the ions involved a  $\text{NF}_3$  b  $\text{BaO}$  c  $\text{NH}_4\text{Cl}$  d  $\text{SrH}_2$  e  $\text{PO}_4^{3-}$  f  $\text{Na}_2\text{O}$  answer a answer b  
answer c answer d answer e answer f problem 4 3 1 4 4 3 1 4

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and nothing else so it is pure when we mix two different pure

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name a molecular compound to name a compound its the prefix name of first element prefix and name of the second element  
with the suffix ide the prefixes are as follows mono 1 di 2 tri 3 tetra 4

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substances as water  $\text{H}_2\text{O}$   $\text{H}_2\text{O}$  and carbon dioxide  $\text{CO}_2$   $\text{CO}_2$  these compounds are very different from ionic compounds like  
sodium chloride  $\text{NaCl}$   $\text{NaCl}$

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ch 2 ch 3 answer a answer b answer c answer d problem 4 4 1 3 calculate the molecular or formula mass of each of the following a p 4 b h 2 o c ca no 3 2 d ch 3 co 2 h acetic acid