

# REVIEW PROBLEMS – EXAM 1

## CHAPTER 3

The following are review exercises for Exam 1. These exercises are provided for you to practice or test yourself for readiness for the exam. These exercises represent many of the types of problems you would be expected to solve on the exam, but are not meant to represent all possible types of problems that could appear on the exam.

1. What is the maximum number of electrons that can be identified with each of the following sets of quantum numbers? In some cases, the answer may be "none". In such case, explain why "none" is the correct answer.

a)  $n = 4, l = 3, m_l = 1$  **2e**

b)  $n = 6, l = 1, m_l = -1, m_s = -1/2$  **1e**

c)  $n = 3, l = 3, m_l = -3$  **none**

d)  $n = 3$  **18e**

e)  $n = 3, l = 2$  **10e**

f)  $n = 4, l = 1, m_l = -1, m_s = -1/2$  **1e**

g)  $n = 5, l = 0, m_l = 1, m_s = +1/2$  **none**

2. Write the full ground state electron configuration for each element. (spdf notation)

a) N

b) S

c) Co

d) Br



3. Complete. Use noble-gas-core abbreviated ground state notation.

Element

spdf notation

N



Fe



I



Cu



Cr



Orbital Diagram



4. A neutral atom has two electrons with  $n = 1$ , eight electrons with  $n = 2$ , eight electrons with  $n = 3$ , and two electrons with  $n = 4$ . Assuming this element is in its ground state, answer the following questions:

a) atomic number: **20**; b) total number of s electrons: **8e**; c) total number of p electrons: **12e**; d) total number of d electrons: **0**; e) number of valence electrons: **2**; f) is the element a metal, metalloid, or nonmetal?

5. Name the element whose electron configuration is a)  $1s^2 2s^2 2p^6 3s^2 3p^3$ : **P**; b)  $[Ar] 4s^2 3d^8$ : **Zn**.

6. List the first element on the periodic table that satisfies each of the following conditions.

a) a completed set of p orbitals: **Ne**; b) two 4p electrons: **Ge**; c) three unpaired electrons: **N**.

7. Complete. Use noble-gas-core abbreviated ground state notation.

Element

spdf notation

Al



$Al^{3+}$



O



$O^{2-}$



Orbital Diagram



# General Chemistry Problems And Solutions

**Deborah Wiegand**



## General Chemistry Problems And Solutions:

*Problem Exercises for General Chemistry* G.Gilbert Long,Forrest C. Hentz,1986-08-14 A question answer supplement to accompany a 1 year general chemistry course This text is a multiple choice format and can be used with any standard general chemistry text The exercises emphasize the importance of problem solving and equation writing in the style used in general chemistry examinations and professional school aptitude examinations such as the MCAT and VCAT

**Problem-Solving** Raymond Chang,2005 By Brandon J Cruickshank Northern Arizona University and Raymond Chang is a success guide written for use with General Chemistry It aims to help students hone their analytical and problem solving skills by presenting detailed approaches to solving chemical problems Solutions for all of the texts even numbered problems are included *General Chemistry* Nebergall,1976

**The Answer Key: A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success (Volume Two) (First Edition** Rachel Turoscy,2018-08-09 The Answer Key A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success Volume 2 is a concise and accessible textbook that covers the critical information a student needs to understand the basic mathematics used in chemistry courses The book provides easy to understand step by step instructions for solving general chemistry problems The book begins with chapters dedicated to solutions kinetics and liquids solids and phase changes In subsequent chapters the text covers important topics like equilibrium concentrations strong and weak acids and bases the Common Ion Effect and reaction mechanisms It also covers the equilibrium between a solid and its respective ions in a solution as well as the second law of Thermodynamics The text also addresses Gibbs Free Energy equilibrium constants and electrolysis calculations Each chapter contains sample problems and practice problems to help further understanding of how math and chemistry go hand in hand The Answer Key is an excellent resource for any undergraduate course that deals with the basic concepts of general chemistry

Survival Guide to General Chemistry Patrick E. McMahon,Rosemary McMahon,Bohdan Khomtchouk,2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists

through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields *Student Solutions Manual: Ssm Chemistry* Deborah Wiegand,2003 This manual contains complete worked out solutions to all follow up problems and about half of all the chapter problems Each chapter of solutions opens with a summary of the text chapter content and a list of key equations needed to solve the problems **Solving General Chemistry Problems** Robert Nelson Smith,Willis Conway Pierce,1980-01-01 Introduction to Spectral Theory in Hilbert Space Gilbert Helmberg,2008-06-11 This introduction to Hilbert space bounded self adjoint operators the spectrum of an operator and operators spectral decomposition is accessible to readers familiar with analysis and analytic geometry 1969 edition *Principles and Techniques of Applied Mathematics* Bernard Friedman,1990-01-01 Stimulating thought provoking study shows how abstract methods of pure mathematics can be used to systematize problem solving techniques in applied mathematics Topics include methods for solving integral equations finding Green s function for ordinary or partial differential equations and for finding the spectral representation of ordinary differential operators Non-Equilibrium Statistical Mechanics Ilya Prigogine,2017-03-17 Groundbreaking monograph by Nobel Prize winner for researchers and graduate students covers Liouville equation anharmonic solids Brownian motion weakly coupled gases scattering theory and short range forces general kinetic equations more 1962 edition *Electricity and Magnetism* Edson Ruther Peck,2013-01-01 This 1953 classic text for advanced undergraduates has been used by generations of physics majors Requiring only some background in general physics and calculus it offers in depth coverage of the field and features problems at the end of each chapter solutions are available for download at the Dover website

*Introduction to Fluid Dynamics* Edward B. McLeod, Jr.,2016-05-13 Concise unified and logical introduction to study of the basic principles of fluid dynamics emphasizes statement of problems in mathematical language Assumes familiarity with algebra of vector fields 1963 edition Gauge Theory and Variational Principles David Bleecker,2005-12-10 This text provides a framework for describing and organizing the basic forces of nature and the interactions of subatomic particles A detailed and self contained mathematical account of gauge theory it is geared toward beginning graduate students and advanced undergraduates in mathematics and physics This well organized treatment supplements its rigor with intuitive ideas Starting with an examination of principal fiber bundles and connections the text explores curvature particle fields Lagrangians and gauge invariance Lagrange s equation for particle fields and the inhomogeneous field equation Additional topics include free Dirac electron fields interactions calculus on frame bundle and unification of gauge fields and gravitation The text concludes with references a selected bibliography an index of notation and a general index **Nonequilibrium Statistical Thermodynamics** Bernard H. Lavenda,2019-04-17 This book develops in detail the statistical foundations of nonequilibrium thermodynamics based on the mathematical theory of Brownian motion Author Bernard H Lavenda demonstrates that thermodynamic criteria emerge in the limit of small thermal fluctuations and in the Gaussian limit where

means and modes of the distribution coincide His treatment assumes the theory of Brownian motion to be a general and practical model of irreversible processes that are inevitably influenced by random thermal fluctuations This unifying approach permits the extraction of widely applicable principles from the analysis of specific models Arranged by argument rather than theory the text is based on the premises that random thermal fluctuations play a decisive role in governing the evolution of nonequilibrium thermodynamic processes and that they can be viewed as a dynamic superposition of many random events Intended for nonmathematicians working in the areas of nonequilibrium thermodynamics and statistical mechanics this book will also be of interest to chemical physicists condensed matter physicists and readers in the area of nonlinear optics

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**The Functions of Mathematical Physics** Harry Hochstadt, 2012-04-30 A modern classic this clearly written incisive textbook provides a comprehensive detailed survey of the functions of mathematical physics a field of study straddling the somewhat artificial boundary between pure and applied mathematics In the 18th and 19th centuries the theorists who devoted themselves to this field pioneers such as Gauss Euler Fourier Legendre and Bessel were searching for mathematical solutions to physical problems Today although most of the functions have practical applications in areas ranging from the quantum theoretical model of the atom to the vibrating membrane some such as those related to the theory of discontinuous groups still remain of purely mathematical interest Chapters One and Two examine orthogonal polynomials with sections on such topics as the recurrence formula the Christoffel Darboux formula the Weierstrass approximation theorem and the application of Hermite polynomials to quantum mechanics Chapter Three is devoted to the principal properties of the gamma function including asymptotic expansions and Mellin Barnes integrals Chapter Four covers hypergeometric functions including a review of linear differential equations with regular singular points and a general method for finding integral representations Chapters Five and Six are concerned with the Legendre functions and their use in the solutions of Laplace's equation in spherical coordinates as well as problems in an  $n$  dimension setting Chapter Seven deals with confluent hypergeometric functions and Chapter Eight examines at length the most important of these the Bessel functions Chapter Nine covers Hill's equations including the expansion theorems

*Elementary Induction on Abstract Structures* Yiannis N. Moschovakis, 2014-06-10 Well written research monograph recommended for students and professionals interested in model theory and definability theory Easy to use and a pleasure to read Bulletin of the American Mathematical Society 1974 edition

**A User's Guide to Ellipsometry** Harland G. Tompkins, 2013-03-21 This text on optics

for graduate students explains how to determine material properties and parameters for inaccessible substrates and unknown films as well as how to measure extremely thin films Its 14 case studies illustrate concepts and reinforce applications of ellipsometry particularly in relation to the semiconductor industry and to studies involving corrosion and oxide growth A User s Guide to Ellipsometry will enable readers to move beyond limited turn key applications of ellipsometers In addition to its comprehensive discussions of the measurement of film thickness and optical constants in film it also considers the trajectories of the ellipsometric parameters  $\Delta$  and  $\Psi$  and how changes in materials affect parameters This volume also addresses the use of polysilicon a material commonly employed in the microelectronics industry and the effects of substrate roughness Three appendices provide helpful references

*Theory of Flight* Richard von Mises, 2012-04-27 Mises classic avoids the formidable mathematical structure of fluid dynamics while conveying by often unorthodox methods a full understanding of the physical phenomena and mathematical concepts of aeronautical engineering

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