

# THE PERIODIC TABLE OF ELEMENTS

THE PERIODIC TABLE OF ELEMENTS																				
1																	18			
1	<div><div>1</div><div>H</div><div>Hydrogen</div></div>																	2	<div><div>2</div><div>He</div><div>Helium</div></div>	
2	<div><div>3</div><div>Li</div><div>Lithium</div></div>	<div><div>4</div><div>Be</div><div>Beryllium</div></div>																	10	<div><div>10</div><div>Ne</div><div>Neon</div></div>
3	<div><div>11</div><div>Na</div><div>Sodium</div></div>	<div><div>12</div><div>Mg</div><div>Magnesium</div></div>																	18	<div><div>18</div><div>Ar</div><div>Argon</div></div>
4	<div><div>19</div><div>K</div><div>Potassium</div></div>	<div><div>20</div><div>Ca</div><div>Calcium</div></div>	<div><div>21</div><div>Sc</div><div>Scandium</div></div>	<div><div>22</div><div>Ti</div><div>Titanium</div></div>	<div><div>23</div><div>V</div><div>Vanadium</div></div>	<div><div>24</div><div>Cr</div><div>Chromium</div></div>	<div><div>25</div><div>Mn</div><div>Manganese</div></div>	<div><div>26</div><div>Fe</div><div>Iron</div></div>	<div><div>27</div><div>Co</div><div>Cobalt</div></div>	<div><div>28</div><div>Ni</div><div>Nickel</div></div>	<div><div>29</div><div>Cu</div><div>Copper</div></div>	<div><div>30</div><div>Zn</div><div>Zinc</div></div>	<div><div>31</div><div>Ga</div><div>Gallium</div></div>	<div><div>32</div><div>Ge</div><div>Germanium</div></div>	<div><div>33</div><div>As</div><div>Arsenic</div></div>	<div><div>34</div><div>Se</div><div>Selenium</div></div>	<div><div>35</div><div>Br</div><div>Bromine</div></div>	<div><div>36</div><div>Kr</div><div>Krypton</div></div>		
5	<div><div>37</div><div>Rb</div><div>Rubidium</div></div>	<div><div>38</div><div>Sr</div><div>Strontium</div></div>	<div><div>39</div><div>Y</div><div>Yttrium</div></div>	<div><div>40</div><div>Zr</div><div>Zirconium</div></div>	<div><div>41</div><div>Nb</div><div>Niobium</div></div>	<div><div>42</div><div>Mo</div><div>Molybdenum</div></div>	<div><div>43</div><div>Tc</div><div>Technetium</div></div>	<div><div>44</div><div>Ru</div><div>Ruthenium</div></div>	<div><div>45</div><div>Rh</div><div>Rhodium</div></div>	<div><div>46</div><div>Pd</div><div>Palladium</div></div>	<div><div>47</div><div>Ag</div><div>Silver</div></div>	<div><div>48</div><div>Cd</div><div>Cadmium</div></div>	<div><div>49</div><div>In</div><div>Indium</div></div>	<div><div>50</div><div>Sn</div><div>Tin</div></div>	<div><div>51</div><div>Sb</div><div>Antimony</div></div>	<div><div>52</div><div>Te</div><div>Tellurium</div></div>	<div><div>53</div><div>I</div><div>Iodine</div></div>	<div><div>54</div><div>Xe</div><div>Xenon</div></div>		
6	<div><div>55</div><div>Cs</div><div>Cesium</div></div>	<div><div>56</div><div>Ba</div><div>Barium</div></div>	<div><div>57-71</div><div>Lanthanide Series</div></div>	<div><div>72</div><div>Hf</div><div>Hafnium</div></div>	<div><div>73</div><div>Ta</div><div>Tantalum</div></div>	<div><div>74</div><div>W</div><div>Tungsten</div></div>	<div><div>75</div><div>Re</div><div>Rhenium</div></div>	<div><div>76</div><div>Os</div><div>Osmium</div></div>	<div><div>77</div><div>Ir</div><div>Iridium</div></div>	<div><div>78</div><div>Pt</div><div>Platinum</div></div>	<div><div>79</div><div>Au</div><div>Gold</div></div>	<div><div>80</div><div>Hg</div><div>Mercury</div></div>	<div><div>81</div><div>Tl</div><div>Thallium</div></div>	<div><div>82</div><div>Pb</div><div>Lead</div></div>	<div><div>83</div><div>Bi</div><div>Bismuth</div></div>	<div><div>84</div><div>Po</div><div>Polonium</div></div>	<div><div>85</div><div>At</div><div>Astatine</div></div>	<div><div>86</div><div>Rn</div><div>Radon</div></div>		
7	<div><div>87</div><div>Fr</div><div>Francium</div></div>	<div><div>88</div><div>Ra</div><div>Radium</div></div>	<div><div>89-103</div><div>Actinide Series</div></div>	<div><div>104</div><div>Rf</div><div>Rutherfordium</div></div>	<div><div>105</div><div>Db</div><div>Dubnium</div></div>	<div><div>106</div><div>Sg</div><div>Seaborgium</div></div>	<div><div>107</div><div>Bh</div><div>Bohrium</div></div>	<div><div>108</div><div>Hs</div><div>Hassium</div></div>	<div><div>109</div><div>Mt</div><div>Meitnerium</div></div>	<div><div>110</div><div>Ds</div><div>Darmstadtium</div></div>	<div><div>111</div><div>Rg</div><div>Roentgenium</div></div>	<div><div>112</div><div>Cn</div><div>Copernicium</div></div>	<div><div>113</div><div>Nh</div><div>Nihonium</div></div>	<div><div>114</div><div>Fl</div><div>Flerovium</div></div>	<div><div>115</div><div>Mc</div><div>Moscovium</div></div>	<div><div>116</div><div>Lv</div><div>Livermorium</div></div>	<div><div>117</div><div>Ts</div><div>Tennessine</div></div>	<div><div>118</div><div>Og</div><div>Oganesson</div></div>		
Lanthanide Series			<div><div>57-71</div><div>Lanthanide Series</div></div>	<div><div>72</div><div>La</div><div>Lanthanum</div></div>	<div><div>73</div><div>Ce</div><div>Cerium</div></div>	<div><div>74</div><div>Pr</div><div>Praseodymium</div></div>	<div><div>75</div><div>Nd</div><div>Neodymium</div></div>	<div><div>76</div><div>Pm</div><div>Promethium</div></div>	<div><div>77</div><div>Sm</div><div>Samarium</div></div>	<div><div>78</div><div>Eu</div><div>Europium</div></div>	<div><div>79</div><div>Gd</div><div>Gadolinium</div></div>	<div><div>80</div><div>Tb</div><div>Terbium</div></div>	<div><div>81</div><div>Dy</div><div>Dysprosium</div></div>	<div><div>82</div><div>Ho</div><div>Holmium</div></div>	<div><div>83</div><div>Er</div><div>Erbium</div></div>	<div><div>84</div><div>Tm</div><div>Thulium</div></div>	<div><div>85</div><div>Yb</div><div>Ytterbium</div></div>	<div><div>86</div><div>Lu</div><div>Lutetium</div></div>		
Actinide Series			<div><div>89-103</div><div>Actinide Series</div></div>	<div><div>104</div><div>Ac</div><div>Actinium</div></div>	<div><div>105</div><div>Th</div><div>Thorium</div></div>	<div><div>106</div><div>Pa</div><div>Protactinium</div></div>	<div><div>107</div><div>U</div><div>Uranium</div></div>	<div><div>108</div><div>Np</div><div>Neptunium</div></div>	<div><div>109</div><div>Pu</div><div>Plutonium</div></div>	<div><div>110</div><div>Am</div><div>Americium</div></div>	<div><div>111</div><div>Cm</div><div>Curium</div></div>	<div><div>112</div><div>Bk</div><div>Berkelium</div></div>	<div><div>113</div><div>Cf</div><div>Californium</div></div>	<div><div>114</div><div>Es</div><div>Einsteinium</div></div>	<div><div>115</div><div>Fm</div><div>Fermium</div></div>	<div><div>116</div><div>Md</div><div>Mendelevium</div></div>	<div><div>117</div><div>No</div><div>Nobelium</div></div>	<div><div>118</div><div>Lr</div><div>Lawrencium</div></div>		

Atomic Number → 1  
Symbol → H  
Name → Hydrogen  
Average Atomic Mass → 1.008

- Alkali Metals
- Alkaline Earth Metals
- Transition Metals
- Other Metals
- Metalloids
- Non-metals
- Halogens
- Noble Gases
- Lanthanides
- Actinides

# Elements Of Chemical Physics

**RS Peters**



## **Elements Of Chemical Physics:**

## Embracing the Melody of Phrase: An Emotional Symphony within **Elements Of Chemical Physics**

In a world used by displays and the ceaseless chatter of instant communication, the melodic splendor and mental symphony developed by the written word usually diminish in to the backdrop, eclipsed by the persistent sound and interruptions that permeate our lives. Nevertheless, set within the pages of **Elements Of Chemical Physics** a stunning literary treasure full of natural feelings, lies an immersive symphony waiting to be embraced. Crafted by an elegant musician of language, that interesting masterpiece conducts readers on a mental trip, skillfully unraveling the hidden tunes and profound influence resonating within each carefully crafted phrase. Within the depths with this emotional analysis, we shall discover the book is central harmonies, analyze their enthralling publishing style, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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