

Basic Chemistry Study Guide 2

1. Hydrogen bonds

1. Form between two polar molecules **covalent bonds**
2. A hydrogen's δ^+ region is attracted to a δ^- region in another molecule.
3. Form between water molecules (Where are the δ^+ and δ^- regions in a water molecule? **hydrogen bonds, at the top of the symbol**)
4. Each hydrogen bond is very weak (can be easily formed and broken).
5. There are many hydrogen bonds in water. Together, these bonds give water its unique properties that are essential for life
6. One of these properties is the ability to be a solvent for other polar molecules.
7. Hydrogen bonds are depicted in figures as dotted or dashed lines between two atoms.

2. Solutions, solvents, and solutes

1. Water is the most abundant molecule in living organisms, and 60% of human body weight is water.
2. Water is located inside and outside of our cells.
3. Body fluids are solutions
4. Define the terms **solution, solvent, and solute**. **Solvent: a substance capable of dissolving other polar molecules and ionic bonds, solutes: molecules dissolved in the solvent, solution: homogeneous mixture**

3. How substances dissolve in water

1. When a solute dissolves, it becomes surrounded by water molecules.
2. For a solute to dissolve, it must attract water molecules.
3. What kinds of substances attract water molecules? **Sugars, nucleic acids, and amino acids**
4. When a solute dissolves it forms hydrogen bonds with water molecules.
5. Define the terms **ion, cation, and anion**. **ion: atom with a charge, cation: positively charged ion, anions: negatively charged ion**
6. Define the terms **hydrophilic** and **hydrophobic** **Hydrophilic: attracted to water, hydrophobic: afraid of water**
7. What types of molecules are hydrophilic and what types are hydrophobic? **Nonpolar hydrophobic, polar hydrophilic**

Elements Of General And Biological Chemistry And Study Guide

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Elements of General and Biological Chemistry and Study Guide John R. Holum, 1996-07 **Elements of General and Biological Chemistry, Study Guide** John R. Holum, 1991-02-26 Revised condensed and based on the Fourth Edition of Fundamentals of General Organic and Biological Chemistry it explores the molecular basis of life Designed for chemistry students aiming for careers in the professional health care field it covers the nature of matter chemical reactions kinetic theory acids bases and salts introductory organic chemistry hydrocarbons aldehydes and ketones carbohydrates lipids proteins enzymes hormones metabolism nucleic acids radioactivity and health and more Features include margin comments special topics on matters of current interest instructional material and review topics Study Guide to Elements of General and Biological Chemistry John R. Holum, 1972 **Study Guide to Accompany Elements of General and Biological Chemistry** John R. Holum, 1975 Elements of General, Organic, and Biological Chemistry John R. Holum, 1995 Text includes chapter summaries and review exercises **Elements of General Organic and Biological Chemistry with Study Guide Set** Holum, 1999-05 *Study Guide to Accompany Elements of General and Biological Chemistry, Fifth Edition* John R. Holum, 1979 Wie Elements of General and Biological Chemistry 4 Th Edition John R. Holum, 1975-01-01 **Study Guide with Answers to Review Questions for Elements of General and Biological Chemistry, Sixth Edition** John R. Holum, 1983 *Elements of General and Biological Chemistry* John R. Holum, 1972 An introduction to the molecular basis of life **Elements of General Organic and Biological Chemistry** Holum, 1991-01-01 **Study Guide to Accompany Elements of General and Biological Chemistry** John R. Holum, 1983 *Elements of General and Biological Chemistry, Laboratory Manual* John R. Holum, 1991-02-05 Revised condensed and based on the Fourth Edition of Fundamentals of General Organic and Biological Chemistry it explores the molecular basis of life Designed for chemistry students aiming for careers in the professional health care field it covers the nature of matter chemical reactions kinetic theory acids bases and salts introductory organic chemistry hydrocarbons aldehydes and ketones carbohydrates lipids proteins enzymes hormones metabolism nucleic acids radioactivity and health and more Features include margin comments special topics on matters of current interest instructional material and review topics Elements of General and Biological Chemistry Holum, 1994-11-01 **Catalog of Copyright Entries. Third Series** Library of Congress. Copyright Office, 1976 **eBook: General, Organic and Biological Chemistry 2e** SMITH, 2012-02-16 eBook General Organic and Biological Chemistry 2e *Farm Crop Production Technology, Field and Forage Crop and Fruit and Vine Production Options* United States. Office of Education, 1970 **Elements of Chemistry** Robert S. Boikess, Kenneth Breslauer, Edward Edelson, 1986 *The Biological Chemistry of the Elements* J. J. R. Frausto da Silva, Robert Joseph Paton Williams, 2001-08-16 This text describes the functional role of the twenty inorganic elements essential to life in living organisms **Teachers Manual to Accompany Elements of General and Biological Chemistry** John R. Holum, 1987

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