

### Simplest aldehyde




The simplest aldehyde, methanal (HCHO) is commonly known as formaldehyde and is used to preserve biological specimens and embalming human remains. The suffix '-al' denotes an aldehyde.

Methanal (HCHO), commonly known as formaldehyde, is the simplest aldehyde. Methanal is used in vast quantities in the manufacture of plastics. The first plastic, known as Bakelite, was made by using phenol (based on benzene) and formaldehyde.

### Naming aldehydes

Aldehydes are named by replacing the 'e' on the name of the hydrocarbon containing the C=O group with the suffix 'al'. To name an aldehyde:

1. identify the longest carbon chain containing the -CHO group
2. delete the 'e' of the parent alkane and add 'al' (for example, ethane becomes ethanal)
3. number the carbon atom of the main chain from the aldehyde. No number is needed for the aldehyde functional group since it is always at the end of the chain and so is always numbered as 1.
4. Name any alkyl groups as for branched alkanes

Name	Molecular formula	Structural formula	Condensed structural formula
methanal	CH <sub>2</sub> O		CH <sub>2</sub> O
ethanal	C <sub>2</sub> H <sub>4</sub> O		CH <sub>3</sub> CHO
propanal	C <sub>3</sub> H <sub>6</sub> O		CH <sub>3</sub> CH <sub>2</sub> CHO

### Ketones

Carbonyl group in ketones is not at the end

Ketones differ from aldehydes in that the C=O can be located on any carbon except those at the ends of the hydrocarbon chain. The compounds have the general formula:



### Example of ketones

Propane (CH<sub>3</sub>COCH<sub>3</sub>), the simplest ketone, is commonly called acetone. The suffix '-One' denotes a ketone

Examples of ketones are:



Propanone is widely used as a solvent, and is commonly known as acetone.

### Naming and drawing aldehydes and ketones and their isomers

Aldehydes are named by replacing the 'e' on the name of the hydrocarbon containing the C=O group with the suffix 'al'.

1. Identify the longest carbon chain containing the -CHO group
2. Delete the 'e' of the parent alkane and add 'al' (for example, ethane becomes ethanal)

# Form And Function An Organic Chemistry Module

**Jack Hassard**



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