

# Periodic Table of the Elements



11.01

H

Hydrogen

3.04

Li

Lithium

6.94

Be

Beryllium

12.01

Na

Sodium

24.31

Mg

Magnesium

39.10

K

Potassium

40.08

Ca

Calcium

88.11

Rb

Rubidium

132.91

Cs

Cesium

223.02

Fr

Francium

14.01

Si

Silicon

28.09

Al

Aluminum

26.98

Si

Silicon

30.97

P

Phosphorus

32.07

S

Sulfur

35.45

Cl

Chlorine

39.96

Ar

Argon

69.72

Ga

Gallium

72.64

Ge

Germanium

74.92

As

Arsenic

78.96

Se

Selenium

81.90

Br

Bromine

83.90

Kr

Krypton

114.82

In

Indium

118.71

Sn

Tin

121.76

Sb

Antimony

127.60

Te

Tellurium

126.91

I

Iodine

131.29

Xe

Xenon

175.08

Tl

Thallium

183.84

Pb

Lead

186.21

Bi

Bismuth

188.01

Po

Polonium

188.91

At

Astatine

189.00

Rn

Radon

1.01

He

Helium

4.00

Ne

Neon

39.95

Ar

Argon

79.90

Kr

Krypton

131.29

Xe

Xenon

222.02

Rn

Radon

14

28.09

Si

Silicon

atomic number

atomic weight

symbol

name

black

solid

blue

liquid

red

gas

white

synthetically prepared

near ambient temp

alkali metals

alkaline earth metals

transitional metals

other metals

nonmetals

noble gases

LABS



Legend for element states:

- Black: solid
- Blue: liquid
- Red: gas
- White: synthetically prepared
- Grey: not fully known

- alkali metals
- alkaline earth metals
- transitional metals
- other metals
- nonmetals
- noble gases

58 Ce Cerium	59 Pr Praseodymium	60 Nd Neodymium	61 Pm Promethium	62 Sm Samarium	63 Eu Europium	64 Gd Gadolinium	65 Tb Terbium	66 Dy Dysprosium	67 Ho Holmium	68 Er Erbium	69 Tm Thulium	70 Yb Ytterbium	71 Lu Lutetium
90 Th Thorium	91 Pa Protactinium	92 U Uranium	93 Np Neptunium	94 Pu Plutonium	95 Am Americium	96 Cm Curium	97 Bk Berkelium	98 Cf Californium	99 Es Einsteinium	100 Fm Fermium	101 Md Mendelevium	102 No Nobelium	103 Lr Lawrencium

# Elements Physical Science

**Lynn Van Gorp**



## Elements Physical Science:

**The World of Elements and Their Properties** Lynn Van Gorp, 2007-08-03 From compounds to chemical reactions readers will learn all about elements their properties and how they react with other elements in this stunning book that features colorful images and intriguing facts Ionic bonds chemical bonds the Periodic Table of Elements mixtures and solutions are some of the topics that are discussed The accessible glossary and index gives readers the tools they need to better understand the content while a fascinating hands on lab activity will leave readers engaged and excited to learn more

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## **Table of Contents Elements Physical Science**

1. Understanding the eBook Elements Physical Science
  - The Rise of Digital Reading Elements Physical Science
  - Advantages of eBooks Over Traditional Books
2. Identifying Elements Physical Science
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Elements Physical Science
  - User-Friendly Interface
4. Exploring eBook Recommendations from Elements Physical Science
  - Personalized Recommendations

- Elements Physical Science User Reviews and Ratings
- Elements Physical Science and Bestseller Lists
- 5. Accessing Elements Physical Science Free and Paid eBooks
  - Elements Physical Science Public Domain eBooks
  - Elements Physical Science eBook Subscription Services
  - Elements Physical Science Budget-Friendly Options
- 6. Navigating Elements Physical Science eBook Formats
  - ePub, PDF, MOBI, and More
  - Elements Physical Science Compatibility with Devices
  - Elements Physical Science Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Elements Physical Science
  - Highlighting and Note-Taking Elements Physical Science
  - Interactive Elements Elements Physical Science
- 8. Staying Engaged with Elements Physical Science
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Elements Physical Science
- 9. Balancing eBooks and Physical Books Elements Physical Science
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Elements Physical Science
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Elements Physical Science
  - Setting Reading Goals Elements Physical Science
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Elements Physical Science
  - Fact-Checking eBook Content of Elements Physical Science

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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