

Periodic Chart of the Elements

IA

THE SYMBOL. Shown in the middle of each block directly below the name of the element. The colour used indicates the physical state of the element under ordinary conditions: black for solids, green for liquids and blue for gases.

THE ATOMIC WEIGHT. Shown below the symbol for each element is the atomic weight as shown in black. The values are taken from the official IUPAC Atomic Weights Of The Elements (2013). For elements not listed in the Report the mass number of the longest lived isotope is shown in brackets.

THE ATOMIC NUMBER. Shown in red in the upper left hand corner.

ELECTRONIC CONFIGURATION. Shown at the upper right as a group of black numbers. When read downwards they indicate the number of electrons normally found in successive energy levels.

NOBLE GASES

2
He
4.003

IIIB IVB VB VIB VIIB

5 B 6 C 7 N 8 O 9 F 10 Ne
13 Al 14 Si 15 P 16 S 17 Cl 18 Ar

19 K 20 Ca 21 Sc 22 Ti 23 V 24 Cr 25 Mn 26 Fe 27 Co 28 Ni 29 Cu 30 Zn 31 Ga 32 Ge 33 As 34 Se 35 Br 36 Kr
37 Rb 38 Sr 39 Y 40 Zr 41 Nb 42 Mo 43 Tc 44 Ru 45 Rh 46 Pd 47 Ag 48 Cd 49 In 50 Sn 51 Sb 52 Te 53 I 54 Xe
55 Cs 56 Ba 57 La 58 Hf 59 Ta 60 W 61 Re 62 Os 63 Ir 64 Pt 65 Au 66 Hg 67 Tl 68 Pb 69 Bi 70 Po 71 At 72 Rn
87 Fr 88 Ra 89 Ac 104 Unq 105 Unp 106 Unh 107 Uns

VIII A

IB IIB

IIIA IVA VA VIA VIIA

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ELECTRONIC CONFIGURATION. Shown at the upper right as a group of black numbers. When read downward they indicate the number of electrons normally found in successive energy levels.

58	59	60	61	62	63	64	65	66	67	68	69	70	71
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu

90	91	92	93	94	95	96	97	98	99	100	101	102	103
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

Elements Of Old English

**Samuel Moore, Thomas a 1880-1945
Knott**

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