peratures at which the composites may be used will be further raised by the use of glasses or metals as matrix materials [24, 25]. Carbon fiber reinforced carbons are stable up to 3000 °C in inert or reducing atmospheres.

[54] D. H. Sowen, F. A. J. Sambell, K. A. D. Lambe, and N. J. Marringley, DOS 1923009 (1968), U.K. A. E.A.
[55] A. A. Behrr and S. J. Bache, Brit. Part. 1177303 (1968).
Rolla Royce.

However, their production is still in the experimental stage.

The further development of these new fiber-reinforced high-temperature materials will open up wide fields of application for carbon fibers.

> Received: March II, 1970 [A 754 IE] German version: Angew. Chem. 52, 400 (1970) Translated by Express Translation in Service, London

Electroplating of Plastics in Theory and Practice

By Kurt Heymann, Wolfgang Riedel, and Günser Woldt (*)

Electroplated plastics combine many advantages of plastics and of metals; they have the law weight and case of shaping of plastics, together with the laster, hardness, and electrical conductivity of metals. An important part of any process for the electroplating of plastics is a pretreatment to ensure good adhesion of the metal film. Etch activation in the straight-through method requires only six steps for pretreatment.

1. Introduction

The rapid growth of the consumption of plastics in the last 20 years is primarily due to the ease with which they can be shaped to obtain useful objects. This was greatly appreciated by industrial users, who required versatility of types and shapes with no great increase in production costs, such as arises e.g. in the processing of metals.

For this reason the users accepted disadvantages in color matching, the attraction of dust by static electricity, and the relatively soft surface of injection molded thermoplastics.

These "made-to-measure materials" offer considerable scope for the work of the chemist. The polymers can be suited to a wide range of uses by modification of the nature of the monomers, the polymerization conditions, and pigmentation, and because of the possibility of reinforcement with glass fibers. Nevertheless, plastics have to compete with the traditional metal.

It seemed natural, therefore, to seek a good combination of plastic and metal, i.e. a genuine composite of plastic and metal.

An important factor here is undeniably the wish to give a plastic surface a metallic luster for decorative reasons, since large sections of the market like the glitter of metals. It is in fact impossible to tell by appearance whether a chromium-plated object is made of metal or plastic; the difference can be felt only by touch. Electroplated plastic feels warm and is much lighter. Because of its lightness, it is easier to transport and handle, and this is a further argument in favor of metallized plastics.

Plastics can be shaped more easily, more cheaply, and in greater variety than metals. To obtain shiny surfaces, the mold into which the plastic is injected is polished instead of each individual article. Without knowing the market, one might think that the electroplating of plastics is a passing fashion. However, this practice has many technical advantages, apart from its decorative value.

Many plastics are damaged by ultraviolet radiation. This damage is prevented by a film of metal. The flammability of plastics is also reduced by metalization.

Use is often made in the electrical industry of chassis having complicated shapes. To produce a metal chassis of this kind, several individual parts have to be fitted together and welded. It has been found in

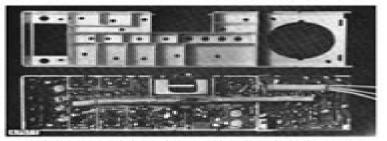


Fig. 1. Charms of ABS plantic (Beach Elektropik Gerbit, Berlin).

^[*] Dr. K. Heymann, Dipt.-Phys. W. Riedel, and Dr. G. Woldt Schering AG, Galvanotechnik.

¹ Berlin 65, Postfach 59 (Germany)

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