

6. Periodic Classification of Elements

Doubt

* The radioisotopes ^{238}Th and ^{234}Pa are successive β^- -decays followed by one α -decay. The atomic numbers and the mass numbers respectively of the resulting radioisotopes are:

(A) 90 and 234

(B) 92 and 234

(C) 92 and 230

(D) 92 and 236

* Consider the following radioactive decay:

(I) $^{238}_{90}\text{Th} \rightarrow ^{234}_{90}\text{Th} + \alpha$

(II) $^{234}_{90}\text{Th} \rightarrow ^{230}_{88}\text{Ra} + \alpha$

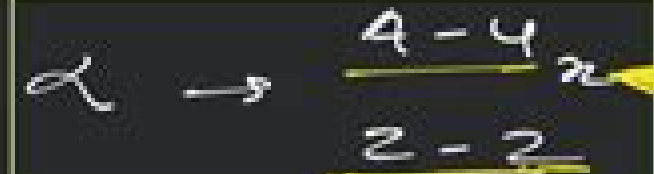
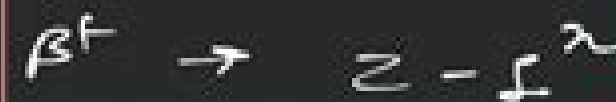
In which case group of parent and daughter elements remains unchanged

(A) In (I)

(B) In (II)

(C) Both in (I) and (II)

(D) None of the above



Elements Of Doubt

JA Banks



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Table of Contents Elements Of Doubt

1. Understanding the eBook Elements Of Doubt
 - The Rise of Digital Reading Elements Of Doubt
 - Advantages of eBooks Over Traditional Books
2. Identifying Elements Of Doubt
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Elements Of Doubt
 - User-Friendly Interface
4. Exploring eBook Recommendations from Elements Of Doubt
 - Personalized Recommendations
 - Elements Of Doubt User Reviews and Ratings
 - Elements Of Doubt and Bestseller Lists
5. Accessing Elements Of Doubt Free and Paid eBooks
 - Elements Of Doubt Public Domain eBooks
 - Elements Of Doubt eBook Subscription Services
 - Elements Of Doubt Budget-Friendly Options
6. Navigating Elements Of Doubt eBook Formats
 - ePub, PDF, MOBI, and More
 - Elements Of Doubt Compatibility with Devices
 - Elements Of Doubt Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Elements Of Doubt
 - Highlighting and Note-Taking Elements Of Doubt
 - Interactive Elements Elements Of Doubt
8. Staying Engaged with Elements Of Doubt

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Elements Of Doubt
- 9. Balancing eBooks and Physical Books Elements Of Doubt
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Elements Of Doubt
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Elements Of Doubt
 - Setting Reading Goals Elements Of Doubt
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Elements Of Doubt
 - Fact-Checking eBook Content of Elements Of Doubt
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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