

Your Name _____

CHM 111

Exam 1

Dr. Look

key

INSTRUCTIONS

⇒ Partial credit will be given for some problems. An attempt must include some original work—more than simply copying information from the problem or listing formulas. Please clearly mark your final answers

AND clearly show all work. Points may not be awarded if work isn't shown—even if the correct solution is given. Always include units.

⇒ Monitor your progress against the time. This is a strictly 50-minute examination—**TIME IS PART OF THE EXAM.** Budget it wisely.

⇒ Ask the instructor for clarifications on the problems whenever necessary. The instructor can sometimes provide more information, but cannot comment on any of your solution work during the exam period.

⇒ You may use a calculator and an appropriate writing implement. There is a Periodic Table on the next page of this exam. Please complete all of your work on the examination pages, including the backs of pages as necessary. Clearly label where the work is located if it is not obvious.

Common SI Prefixes

10^9	giga (G)
10^6	mega (M)
10^3	kilo (k)
10^1	deci (d)
10^{-2}	centi (c)
10^{-3}	milli (m)
10^{-6}	micro (μ)
10^{-9}	nano (n)
10^{-12}	pico (p)

Thus, $1 \text{ cm} = 10^{-2} \text{ m}$.

$1 \text{ mL} = 1 \text{ cm}^3$

$D = m/V$

$$V = l \times w \times h$$

$$N_A = 6.022 \times 10^{23} \text{ molecules, atoms, peas, etc}$$

$$\text{Atomic mass} = \sum(\text{fraction } n \times \text{mass of } n)$$

$$\text{Individual deviation} = |x - \bar{x}|$$

$$\text{Avg. Deviation} = \frac{\sum_{i=1}^n (|x_i - \bar{x}|)}{n}$$

Examinations Solutions For Introductory Chemistry

Joacim Rocklöv



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