

6. Periodic Classification of Elements

Doubt

* The radioisotopes ^{234}Th and ^{234}Pa are successive β^- -decays followed by one α -decay. The atomic numbers and the mass numbers respectively of the resulting radioisotope are:

(A) 90 and 230

(B) 92 and 230

(C) 92 and 234

(D) 92 and 236

* Consider the following radioactive decay:

(I) $^{90}_{40}\text{Th} \rightarrow ^{90}_{41}\text{Pa}$ and

(II) $^{90}_{40}\text{Th} \rightarrow ^{86}_{82}\text{Pb}$

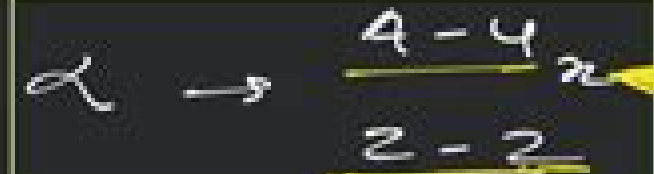
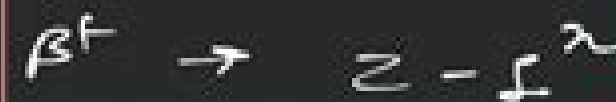
In which case group of parent and daughter elements remains unchanged

(A) In (I)

(B) In (II)

(C) Both in (I) and (II)

(D) None of the above



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