

THE PERIODIC TABLE OF ELEMENTS

1

THE PERIODIC TABLE OF ELEMENTS

18

1																	18			
1	<div><div>1.008</div><div>H</div><div>Hydrogen</div></div>																	2	<div><div>4.003</div><div>He</div><div>Helium</div></div>	
2	<div><div>6.941</div><div>Li</div><div>Lithium</div></div>	<div><div>9.012</div><div>Be</div><div>Beryllium</div></div>																	10	<div><div>20.18</div><div>Ne</div><div>Neon</div></div>
3	<div><div>22.990</div><div>Na</div><div>Sodium</div></div>	<div><div>24.305</div><div>Mg</div><div>Magnesium</div></div>																	18	<div><div>39.95</div><div>Ar</div><div>Argon</div></div>
4	<div><div>39.098</div><div>K</div><div>Potassium</div></div>	<div><div>40.078</div><div>Ca</div><div>Calcium</div></div>	<div><div>44.956</div><div>Sc</div><div>Scandium</div></div>	<div><div>47.867</div><div>Ti</div><div>Titanium</div></div>	<div><div>50.942</div><div>V</div><div>Vanadium</div></div>	<div><div>51.996</div><div>Cr</div><div>Chromium</div></div>	<div><div>54.938</div><div>Mn</div><div>Manganese</div></div>	<div><div>55.935</div><div>Fe</div><div>Iron</div></div>	<div><div>58.933</div><div>Co</div><div>Cobalt</div></div>	<div><div>58.933</div><div>Ni</div><div>Nickel</div></div>	<div><div>63.546</div><div>Cu</div><div>Copper</div></div>	<div><div>65.38</div><div>Zn</div><div>Zinc</div></div>	<div><div>69.723</div><div>Ga</div><div>Gallium</div></div>	<div><div>72.64</div><div>Ge</div><div>Germanium</div></div>	<div><div>74.922</div><div>As</div><div>Arsenic</div></div>	<div><div>78.971</div><div>Se</div><div>Selenium</div></div>	<div><div>79.904</div><div>Br</div><div>Bromine</div></div>	<div><div>83.80</div><div>Kr</div><div>Krypton</div></div>		
5	<div><div>85.468</div><div>Rb</div><div>Rubidium</div></div>	<div><div>87.62</div><div>Sr</div><div>Strontium</div></div>	<div><div>88.906</div><div>Y</div><div>Yttrium</div></div>	<div><div>91.224</div><div>Zr</div><div>Zirconium</div></div>	<div><div>92.906</div><div>Nb</div><div>Niobium</div></div>	<div><div>95.94</div><div>Mo</div><div>Molybdenum</div></div>	<div><div>98</div><div>Tc</div><div>Technetium</div></div>	<div><div>101.07</div><div>Ru</div><div>Ruthenium</div></div>	<div><div>102.905</div><div>Rh</div><div>Rhodium</div></div>	<div><div>106.42</div><div>Pd</div><div>Palladium</div></div>	<div><div>107.868</div><div>Ag</div><div>Silver</div></div>	<div><div>112.411</div><div>Cd</div><div>Cadmium</div></div>	<div><div>114.818</div><div>In</div><div>Indium</div></div>	<div><div>117.904</div><div>Sn</div><div>Tin</div></div>	<div><div>121.757</div><div>Sb</div><div>Antimony</div></div>	<div><div>127.6</div><div>Te</div><div>Tellurium</div></div>	<div><div>126.905</div><div>I</div><div>Iodine</div></div>	<div><div>131.29</div><div>Xe</div><div>Xenon</div></div>		
6	<div><div>132.905</div><div>Cs</div><div>Cesium</div></div>	<div><div>137.327</div><div>Ba</div><div>Barium</div></div>	<div><div>138.905</div><div>La</div><div>Lanthanum</div></div>	<div><div>178.48</div><div>Hf</div><div>Hafnium</div></div>	<div><div>180.948</div><div>Ta</div><div>Tantalum</div></div>	<div><div>183.84</div><div>W</div><div>Tungsten</div></div>	<div><div>186.207</div><div>Re</div><div>Rhenium</div></div>	<div><div>188.906</div><div>Os</div><div>Osmium</div></div>	<div><div>190.224</div><div>Ir</div><div>Iridium</div></div>	<div><div>195.084</div><div>Pt</div><div>Platinum</div></div>	<div><div>196.967</div><div>Au</div><div>Gold</div></div>	<div><div>200.59</div><div>Hg</div><div>Mercury</div></div>	<div><div>204.38</div><div>Tl</div><div>Thallium</div></div>	<div><div>207.2</div><div>Pb</div><div>Lead</div></div>	<div><div>208.98</div><div>Bi</div><div>Bismuth</div></div>	<div><div>209</div><div>Po</div><div>Polonium</div></div>	<div><div>210</div><div>At</div><div>Astatine</div></div>	<div><div>222</div><div>Rn</div><div>Radon</div></div>		
7	<div><div>223</div><div>Fr</div><div>Francium</div></div>	<div><div>226</div><div>Ra</div><div>Radium</div></div>	<div><div>227</div><div>Ac</div><div>Actinium</div></div>	<div><div>261</div><div>Rf</div><div>Rutherfordium</div></div>	<div><div>262</div><div>Db</div><div>Dubnium</div></div>	<div><div>263</div><div>Sg</div><div>Seaborgium</div></div>	<div><div>264</div><div>Bh</div><div>Bohrium</div></div>	<div><div>277</div><div>Hs</div><div>Hassium</div></div>	<div><div>285</div><div>Mt</div><div>Meitnerium</div></div>	<div><div>289</div><div>Ds</div><div>Darmstadtium</div></div>	<div><div>290</div><div>Rg</div><div>Roentgenium</div></div>	<div><div>291</div><div>Cn</div><div>Copernicium</div></div>	<div><div>294</div><div>Nh</div><div>Nihonium</div></div>	<div><div>299</div><div>Fl</div><div>Flerovium</div></div>	<div><div>301</div><div>Mc</div><div>Moscovium</div></div>	<div><div>304</div><div>Lv</div><div>Livermorium</div></div>	<div><div>315</div><div>Ts</div><div>Tennessine</div></div>	<div><div>318</div><div>Og</div><div>Oganesson</div></div>		

Atomic Number →

1

1.008

← Average Atomic Mass

Symbol →

H

Name →

Hydrogen

Alkali Metals

Alkaline Earth Metals

Transition Metals

Other Metals

Metalloids

Non-metals

Halogens

Noble Gases

Lanthanides

Actinides

Atomic Number → 1 1.008 ← Average Atomic Mass

Symbol → **H**

Name → Hydrogen

- Alkali Metals
- Alkaline Earth Metals
- Transition Metals
- Other Metals
- Metalloids
- Non-metals
- Halogens
- Noble Gases
- Lanthanides
- Actinides

• Lanthanide Series

•• Actinide Series

La Lanthanum	Ce Cerium	Pr Praseodymium	Nd Neodymium	Pm Promethium	Sm Samarium	Eu Europium	Gd Gadolinium	Tb Terbium	Dy Dysprosium	Ho Holmium	Er Erbium	Tm Thulium	Yb Ytterbium	Lu Lutetium
Ac Actinium	Th Thorium	Pa Protactinium	U Uranium	Np Neptunium	Pu Plutonium	Am Americium	Cm Curium	Bk Berkelium	Cf Californium	Es Einsteinium	Fm Fermium	Md Mendelevium	No Nobelium	Lr Lawrencium

Elements Of Chemical Physics

Ying-Ying Zheng



Elements Of Chemical Physics:

This book delves into Elements Of Chemical Physics. Elements Of Chemical Physics is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Elements Of Chemical Physics, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Elements Of Chemical Physics
- Chapter 2: Essential Elements of Elements Of Chemical Physics
- Chapter 3: Elements Of Chemical Physics in Everyday Life
- Chapter 4: Elements Of Chemical Physics in Specific Contexts
- Chapter 5: Conclusion

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4. In chapter 3, this book will examine the practical applications of Elements Of Chemical Physics in daily life. The third chapter will showcase real-world examples of how Elements Of Chemical Physics can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Elements Of Chemical Physics in specific contexts. This chapter will explore how Elements Of Chemical Physics is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Elements Of Chemical Physics. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Elements Of Chemical Physics.

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