



# Funny Convulsing And Funny Confusing

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web jan 13 2023 the series is the row below the lanthanide series which is located underneath the main body of the periodic table lanthanide and actinide series are both referred to as rare earth metals these elements all have a high diversity in oxidation numbers all of the actinides are radioactive

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web on the basis of the entrance of the last electron either into 4 f orbital or 5 f orbital these elements are classified into two categories called lanthanides and actinides respectively lanthanides the name lanthanides comes from lanthanum la z 57 because these elements are closely related to lanthanum

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web the lanthanides and actinides including group 3 metals due to their chemical similarities account for around a third of the periodic table and they are key to a wide range of basic science research and societally beneficial technologies since 2019 is designated as international year of the periodic table it is an appropriate time to

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web the actinides are elements 89 to 103 and fill their 5f sublevel progressively actinides are typical metals and have properties of both the d block and the f block elements but they are also radioactive lanthanides have different chemistry from transition metals because their 4f orbitals are shielded from the atom's environment

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and the actinides are formed while filling the 5f sublevel generally speaking the lanthanides

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the most important part of the male reproductive system is to produce the sperms gametes and transfer it to the female reproductive organ to get fertilized the male reproductive organ consists mainly of scrotum and penis and lay externally to the body and around the pelvic girdle this is because the see more

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