



Education Of Bear Hunter

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web we can balance it for charge by adding two electrons to the right side of the equation so that the net charge on each side is 0 oxidation $\text{Ni} \rightarrow \text{Ni}^{2+} + 2\text{e}^-$ now that the oxidation half reaction is balanced it tells us that two electrons are produced for every atom of nickel oxidized but where do those electrons go

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web in the redox equation below that occurs in our stomach which of the following pairs identifies the reducing and oxidizing agents respectively c₆h₈o₆ 2h₂no₂ c₆h₆o₆ 2h₂o 2no choose 1 answer c₆h₈o₆ and no₂ a c₆h₈o₆ and no₂ no₂ and c₆h₈o₆ b no₂ and c₆h₈o₆ h and no₂ c h and

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