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# Health Care in Uganda

## *Selected Issues*

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*in collaboration with*  
*Dominic Habte and Mary Mubona*

# Health Care In Uganda Selected Issues

**Thirumalaisamy P.  
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 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \quad \Delta h = 71.2 \text{ kJ}$   
 $\text{C}_2\text{H}_6(\text{g}) \rightarrow \frac{7}{2} \text{O}_2(\text{g}) + 2 \text{H}_2\text{O}(\text{g}) \quad \Delta h = -1560 \text{ kJ}$



co<sub>2</sub> g 3 h 2 o g  $\Delta h$  283 kj g 1 c<sub>2</sub>h<sub>6</sub> g

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web thus for reaction 9 2 in which 1 mole of  $\text{H}_3\text{O}^+$  and 1 mole of  $\text{OH}^-$  combine to form 2 moles of  $\text{H}_2\text{O}$   $\Delta h = 55.8 \text{ kJ}$  as shown note that the enthalpy change  $\Delta h_{\text{products}} - \Delta h_{\text{reactants}}$  is positive if heat is absorbed that is if  $\Delta h_{\text{products}}$

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