# Effective Reading

TEACHER'S BOOK

Simon Greenall and Michael Swan

Cambridge University Press



# **Effective Reading Reading Skills For Advanced Students**

**Karen Gazith** 

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web stant is the foundation of thermochemistry it allows us to describe heat flow in terms of the change in a state function enthalpy change is path independent note with regard to choice b that enthalpy change equals internal energy change only if the volume of the system remains constant and thus no thermody namic work is performed 5 b

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web 1 answer a explanation the value of  $\Delta h$  is negative therefore the reaction is exothermic or heat releasing 2 answer a 1 explanation the entire reaction was multiplied by 2 therefore  $\Delta h$  must also be halved 3 answer c explanation option c is incorrect because 566 kj mol is the  $\Delta h$  per mole of o2 and not for 2 moles

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web we will revisit this today and go on to see that if we use a special kind of thermochemical reaction called the standard enthalpy of formation we can calculate enthalpies of reactions without having to manipulate a series of individual thermochemical equations for each step

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