

Half Reaction

Standard Potential (V)

↑ stronger oxidizing agent	F₂	+	2e ⁻	⇌	2F ⁻	+2.87
	Pb⁴⁺	+	2e ⁻	⇌	Pb ²⁺	+1.67
	Cl₂	+	2e ⁻	⇌	2Cl ⁻	+1.36
	O ₂ + 4H ⁺	+	4e ⁻	⇌	2H ₂ O	+1.23
	Ag ⁺	+	1e ⁻	⇌	Ag	+0.80
	Fe ³⁺	+	1e ⁻	⇌	Fe ²⁺	+0.77
	Cu ²⁺	+	2e ⁻	⇌	Cu	+0.34
	2H ⁺	+	2e ⁻	⇌	H ₂	0.00
	Pb ²⁺	+	2e ⁻	⇌	Pb	-0.13
	Fe ²⁺	+	2e ⁻	⇌	Fe	-0.44
	Zn ²⁺	+	2e ⁻	⇌	Zn	-0.76
	Al ³⁺	+	3e ⁻	⇌	Al	-1.66
	Mg ²⁺	+	2e ⁻	⇌	Mg	-2.36
	Li ⁺	+	1e ⁻	⇌	Li	-3.05

↓
stronger reducing agent

Electrode Potentials

Brendan G. Carr

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Electrode Potentials:

Electrode Potentials Richard G. Compton, 2024

Handbook of Electrochemistry Cynthia G. Zoski, 2007-02-07

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Tables of Standard Electrode Potentials Giulio Milazzo, Sergio Caroli, V. K. Sharma, 1978 The Electrode Potential and Replacing Power of Manganese and Study of Some of Its Alloys Harvey Day Royce, 1926 **Fundamentals of Electrochemistry** Vladimir S. Bagotsky, 2005-12-02

Fundamentals of Electrochemistry provides the basic outline of most topics of theoretical and applied electrochemistry for students not yet familiar with this field as well as an outline of recent and advanced developments in electrochemistry for people who are already dealing with electrochemical problems The content of this edition is arranged so that all basic information is contained in the first part of the book which is now rewritten and simplified in order to make it more accessible and used as a textbook for undergraduate students More advanced topics of interest for postgraduate levels come in the subsequent parts This updated second edition focuses on experimental techniques including a comprehensive chapter on physical methods for the investigation of electrode surfaces New chapters deal with recent trends in electrochemistry including nano and micro electrochemistry solid state electrochemistry and electrocatalysis In addition the authors take into account the worldwide renewal of interest for the problem of fuel cells and include chapters on batteries fuel cells and double layer capacitors

Physical Chemistry Robert J. Silbey, Robert A. Alberty, George A. Papadantonakis, Mouni G. Bawendi, 2022-06-15 Ever since Physical Chemistry was first published in 1913 it has remained a highly effective and relevant learning tool thanks to the efforts of physical chemists from all over the world Each new edition has benefited from their suggestions and expert advice The result of this remarkable tradition is now in your hands Introduction to Hydrogen Technology K. S. V. Santhanam, Roman J. Press, Massoud J. Miri, Alla V. Bailey, Gerald A. Takacs, 2017-09-19 Introduces the field of hydrogen technology and explains the basic chemistry underlying promising and innovative new technologies This new and completely updated edition of Introduction to Hydrogen Technology explains at an introductory level the scientific and technical aspects of hydrogen technology It incorporates information on the latest developments and the current research in the field including new techniques for isolating and storing hydrogen usage as a fuel for automobiles residential power systems mobile power systems and space applications Introduction to Hydrogen Technology Second Edition features

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