

Table 8.1: Fundamental physical formats, constants, and relationships between the units of the elementary quantities count, amount, and charge.

Format	Name	Definitions	Numerical value	Unit *
\underline{N}	count, molecular #	count $N_X = N \cdot U_X$	\triangle MU = x	
\underline{n}	amount, molar	amount $n_X = N_X \cdot N_A^{-1}$	\square MU = mol	
\underline{q}	charge, electrical	charge $Q_{eX} = n_X \cdot z_X \cdot F$	\circ MU = C	
\underline{N}	Boltzmann constant §	$k = f \cdot e = R / N_A = 1.380\,649 \cdot 10^{-23}$	$\text{J} \cdot \text{x}^{-1} \cdot \text{K}^{-1}$	
\underline{n}	gas constant	$R = f \cdot F = k \cdot N_A = 8.314\,462\,618$	$\text{J} \cdot \text{mol}^{-1} \cdot \text{K}^{-1}$	
\underline{q}	electromotive constant §	$f = k / e = R / F = 8.617\,333\,262 \cdot 10^{-5}$	$\text{J} \cdot \text{C}^{-1} \cdot \text{K}^{-1}$	
$\underline{N} / \underline{n}$	Avogadro constant §	$N_A = R / k = F / e = 6.022\,140\,76 \cdot 10^{23}$	$\text{x} \cdot \text{mol}^{-1}$	
$\underline{q} / \underline{N}$	elementary charge §,®	$e = k / f = F / N_A = 1.602\,176\,634 \cdot 10^{-19}$	$\text{C} \cdot \text{x}^{-1}$	
$\underline{q} / \underline{n}$	Faraday constant	$F = R / f = e \cdot N_A = 96\,485.332\,12$	$\text{C} \cdot \text{mol}^{-1}$	

* The motive quantity with motive unit MU defines the physical format.

® Count N_X is the number N of elementary entities U_X with the abstract elementary unit [x] (Gnaiger 2020).

§ Redefinition of SI base units came into force on 2019-05-20; Bureau International des Poids et Mesures (2019) The International System of Units (SI), 9th edition.

§ A name or symbol was not found in the literature for the electromotive constant f introduced here.

® Elementary charge $e \triangleq Q_{dp^+} / N_{p^+} = Q_{dp^+}$ is charge per proton count or charge per elementary proton U_{p^+} .

Electrochemical Constants

Mike Jess

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

Electrochemical Constants:

Reviewing **Electrochemical Constants**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Electrochemical Constants**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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