

Common flexible working arrangements

1



Flextime

2



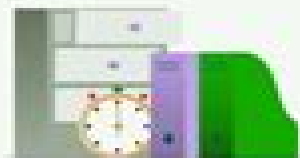
Part-time position

3



Remote work

4



Compressed
workweek

5



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6



Job-sharing

Flexible Work Arrangements

Puja Mehta



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$$3\text{CO} + 7\text{H}_2 \rightarrow \text{C}_3\text{H}_8 + 3\text{H}_2\text{O}$$

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g al x 1 mol al x 2 mol al₂O₃ x 101 964 g al₂O₃ 302 3 g al₂O₃ 26 982 g al 4 mol al 1 mol al₂O₃ c determine the percent yield
 yield actual x 100 260 0 g x 100 86 01 theoretical 302 3 g

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web 7 na 2 c 2 o 4 is the limiting reactant percent yield 86 6 8 only four molecules can be made 9 this amount cannot be
 weighted by ordinary balances and is worthless 10 nitrogen is the limiting reagent 11 yes methane is the limiting reagent 12
 c is the limiting reagent 4 33 g of h 2 are left over 13

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 yield theoretical yield actual yield 100 based on this definition we would expect a percent yield

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80 g I_2O_5 1 mol I_2O_5 1 mol I_2 \times 1 333 8 g I_2O_5 1 mol I_2O_5 28 g CO 1 mol CO

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