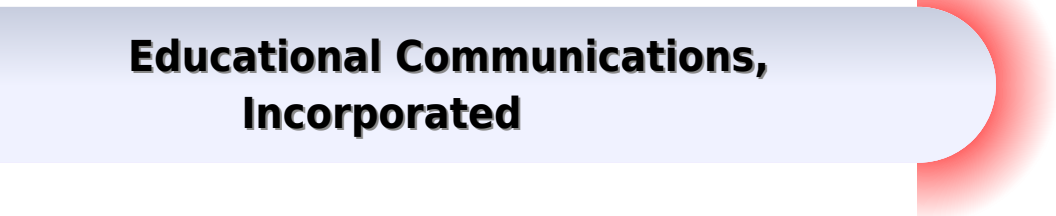




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these are the two types of bonds out of which every compound is made of 2 example of compounds example of compounds
includes water H_2O hydrogen peroxide H_2O_2 etc

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water are composed of molecules that is they are molecular compounds sodium chloride on the other hand contains ions it is
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molecular compounds covalent compounds result when atoms share rather than transfer gain or lose electrons covalent
bonding is an important and extensive concept in chemistry and it will be treated in considerable

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it is ionic write the symbols for the ions involved a NF_3 b BaO c NH_4Cl d SrH_2 e PO_4^{3-} f Na_2O answer a answer b
answer c answer d answer e answer f problem 4 3 1 4 4 3 1 4

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and nothing else so it is pure when we mix two different pure

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with the suffix ide the prefixes are as follows mono 1 di 2 tri 3 tetra 4

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substances as water H_2O H_2O and carbon dioxide CO_2 CO_2 these compounds are very different from ionic compounds like
sodium chloride NaCl NaCl

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ch 2 ch 3 answer a answer b answer c answer d problem 4 4 1 3 calculate the molecular or formula mass of each of the following a p 4 b h 2 o c ca no 3 2 d ch 3 co 2 h acetic acid