

15.6: The effect of Temperature on Reaction rate

the rates of chemical reactions are, in general, highly sensitive to temperature. $1^\circ \text{C} \approx 1.5$, $10^\circ \text{C} \approx 2.5$

The Arrhenius Equation

- the temperature dependence of the reaction rate is contained in the rate constant k (only constant when temp remains c).
- the modern form of the Arrhenius equation shows the relationship between the rate constant (k) & the temp in k (T)

$$K = A e^{-\frac{E_a}{RT}}$$

FREQUENCY factor exponential factor activation E

Activation energy (E_a): an energy barrier or hump that must be surmounted in order for the reactants to transform into products.

Frequency factor (A): the # of times that the reactants approach the activation barrier per unit time.



THE ACTIVATION ENERGY

- to get from the reactant to the product, the molecule must go through a high-energy state called the activated complex. This is because energy is required to initially weaken the $\text{H}_3\text{C}-\text{N}$ bond & allow the NC group to begin to rotate.

→ the energy required to reach the activated complex is the activation energy.

$\uparrow E_a \quad \downarrow$ reaction rate (at a given temp.)

THE FREQUENCY FACTOR

- for the reaction, the frequency factor represents the rate at which the NC part of the molecule wags (wags side to side). With each wag, the reactant approaches the activation barrier.

↳ not equivalent to surmounting it.

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Basic Concepts of Chemistry Leo J. Malone, 2000-05 **EBOOK: GENERAL CHEMISTRY, THE ESSENTIAL CONCEPTS** CHANG, 2013-01-07 **EBOOK GENERAL CHEMISTRY THE ESSENTIAL CONCEPTS** *Basic Concepts of Chemistry* Leo J. Malone, Theodore Dolter, 2008-12-03 Engineers who need to have a better understanding of chemistry will benefit from this accessible book. It places a stronger emphasis on outcomes assessment which is the driving force for many of the new features. Each section focuses on the development and assessment of one or two specific objectives. Within each section a specific objective is included, an anticipatory set to orient the reader, content discussion from established authors and guided practice problems for relevant objectives. These features are followed by a set of independent practice problems. The expanded Making it Real feature showcases topics of current interest relating to the subject at hand such as chemical forensics and more medical related topics. Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance. *Key Concepts in Environmental Chemistry* Grady Hanrahan, 2011-08-02 *Key Concepts in Environmental Chemistry* provides a modern and concise introduction to environmental chemistry principles and the dynamic nature of environmental systems. It offers an intense one semester examination of selected concepts encountered in this field of study and provides integrated tools in explaining complex chemical problems of environmental importance. Principles typically covered in more comprehensive textbooks are well integrated into general chapter topics and application areas. The goal of this textbook is to provide students with a valuable resource for learning the basic concepts of environmental chemistry from an easy to follow condensed application and inquiry based perspective. Additional statistical sampling, modeling and data analysis concepts and exercises will be introduced for greater understanding of the underlying processes of complex environmental systems and fundamental chemical principles. Each chapter will have problem oriented exercises with examples throughout the body of the chapter that stress the important concepts covered and research applications case studies from experts in the field. Research applications will be directly tied to theoretical concepts covered in the chapter. Overall, this text provides a condensed and integrated tool for student learning and covers key concepts in the rapidly developing field of environmental chemistry. Intense one semester approach to learning. Application based approach to learning theoretical concepts. In depth analysis of field based and in situ analytical techniques. Introduction to environmental modeling. **Core Concepts in Supramolecular Chemistry and Nanochemistry** Jonathan W. Steed, David R. Turner, Karl Wallace, 2007-04-30 Supramolecular chemistry and nanochemistry are two strongly interrelated cutting edge frontiers in research in the chemical sciences. The results of recent work in the area are now an increasing part of modern degree courses and hugely important to researchers. *Core Concepts in Supramolecular Chemistry and Nanochemistry* clearly outlines the fundamentals that underlie supramolecular chemistry and nanochemistry and takes an umbrella view of the whole area. This concise textbook traces the

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