Solving for Equilibrium Concentration

Step #2: We "ICE" the problem, beginning with the Initial concentrations

$$H_2O(g) + CO(g) = H_2(g) + CO_2(g)$$

<u>I</u> nitial:	8	6	0	0
<u>C</u> hange:	-x	-x	+x	+x
Equilibrium:	8-x	6-x	×	x

Equilibrium Freshman Chemistry Problems And How To Solve Them Part

S Marginson

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