

Electron Configurations in the Periodic Table

1 H 1s																	2 He 1s						
3 Li 2s	4 Be																	5 B 2p	6 C 2p	7 N 2p	8 O 2p	9 F 2p	10 Ne 2p
11 Na 3s	12 Mg																	13 Al 3p	14 Si 3p	15 P 3p	16 S 3p	17 Cl 3p	18 Ar 3p
19 K 4s	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn 3d	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As 4p	34 Se 4p	35 Br	36 Kr	49 In	50 Sn	51 Sb 5p	52 Te 5p	53 I	54 Xe
37 Rb 5s	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc 4d	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb 5p	52 Te 5p	53 I	54 Xe	81 Tl	82 Pb	83 Bi 6p	84 Po 6p	85 At	86 Rn
55 Cs 6s	56 Ba	57 La	72 Hf	73 Ta	74 W	75 Re 5d	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi 6p	84 Po 6p	85 At	86 Rn	113	114				
87 Fr 7s	88 Ra	89 Ac	104 Rf	105 Db	106 Sg	107 Bh 6d	108 Hs	109 Mt	110	111	112	113	114										

58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr

Electronic Structure Properties The Pe

Jin-Ying Zhang



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