

Introduction

The section briefly outlines the context of the report, with clear objectives, expectations and achievements. They must be in continuous pros.

Sections

Should cover what was done in each section, what can be done better, challenges and possible solutions

Discussion

This section involves detailed explanation of the sections and per what was observed and comparing with standard systems or existing literature. It also captures the implication of the suggested solutions.

Conclusion

It captures the main home take message in each section

Acknowledgements

In this section, the author acknowledges those who supported you during the course. Should not be more than 100 words

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It lists all the literature cited in the document

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Guidelines F Report Writing Canadian

Thomas Griffiths



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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414 l 22 414 l we can substitute 101 325 kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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web the properties of water at constant volume can be represented as a function of pressure and temperature as shown in the diagram below figure 1 the pressure temperature relationship at constant volume the saturation point of water is the condition at which a phase change occurs from liquid to vapour or vapour to liquid

6 3 relationships among pressure temperature volume and - Jul 07 2023

web figure pageindex 5 the empirically determined relationships among pressure volume temperature and amount of a gas the thermometer and pressure gauge indicate the temperature and the pressure qualitatively the level in the flask indicates the volume and the number of particles in each flask indicates relative amounts