diffusion equation, elementary

The equation obtained by using Fick's law to relate the current density to the neutron density in a continuity equation.

diffusion length

A length appearing in the description of diffusion when Fick's law applies. It is the square of the diffusion length (called the diffusion wea) equal to one-sixth the mean square distance traveled by a particle of a given type or class from appearance to disappearance within an infinite homogeneous medium.

diffusion, molecular (isotopes separation)

A process whereby one substance gradually interpenetrates another substance as a result of the continuous thermal motion of the individual molecules. Diffusion due to molecular motion satisfies a partial differential equation of the form

$$\frac{\partial \rho}{\partial t} = D \frac{\partial^2 \rho}{\partial x^2}$$

where ρ is the density of the material diffusion, t is the time and x is distance in the direction of the density gradient. The quantity D is known as the coefficient of diffusion for the mixture in question; it turns out to be rate-dotermining in all separation processes using gases or liquids at pressures where molecular collisions are a significant phenomenon.

diffusivity (chemical engineering) The relative rate of flow per unit area of a particular constituent of a mixture divided by the gradient of composition, temperature or other property considered to be causing the diffusion. Also diffusion co-efficient.

diploid (biophysics)

Having two complete sets of chromosomes, the normal somatic number of chromosomes in higher organisms, as opposed to the half number haploid, (q.v.) found in mature germ cells.

(See t)

direct radiation (radiology)

All radiation other than the useful beam emanating from the x-ray tube anode or the inside of the tube or tube container. (See useful beam)

disadvantage factor

The ratio of the average neutron flux in a cell to the average neutron flux within the fissionable material.

discomposition

The process in which an atom is knocked out of its position in a crystal lattice by direct nuclear impact, e.g., by fast neutrons or by fast ions which have been previously knocked out of their lattice positions. The atom so displaced eventually comes to rest at an interstitial position or at a lattice edge.

discomposition effect

Change in chemical or physical properties resultant from discomposition. Also called Wigner effect.

discriminator

(See pulse beight discriminator)

disintegration

A spontaneous nuclear transformation (radioactivity) characterized by the emission of energy from the nucleus. When numbers of nuclei are involved the process is characterized by a definite half-life. (See radioactive balf-life) (Chemistry)

- 1. Radioactive decay.
- The transformation of one nuclide into one or more different nuclides

Glossary Of Terms In Nuclear Science T

A Gutmann

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Table of Contents Glossary Of Terms In Nuclear Science T

- 1. Understanding the eBook Glossary Of Terms In Nuclear Science T
 - The Rise of Digital Reading Glossary Of Terms In Nuclear Science T
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Glossary Of Terms In Nuclear Science T
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Glossary Of Terms In Nuclear Science T
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Glossary Of Terms In Nuclear Science T
 - Personalized Recommendations
 - $\circ\,$ Glossary Of Terms In Nuclear Science T User Reviews and Ratings
 - Glossary Of Terms In Nuclear Science T and Bestseller Lists
- 5. Accessing Glossary Of Terms In Nuclear Science T Free and Paid eBooks
 - Glossary Of Terms In Nuclear Science T Public Domain eBooks
 - Glossary Of Terms In Nuclear Science T eBook Subscription Services
 - Glossary Of Terms In Nuclear Science T Budget-Friendly Options

- 6. Navigating Glossary Of Terms In Nuclear Science T eBook Formats
 - o ePub, PDF, MOBI, and More
 - Glossary Of Terms In Nuclear Science T Compatibility with Devices
 - Glossary Of Terms In Nuclear Science T Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of Glossary Of Terms In Nuclear Science T
 - Highlighting and Note-Taking Glossary Of Terms In Nuclear Science T
 - Interactive Elements Glossary Of Terms In Nuclear Science T
- 8. Staying Engaged with Glossary Of Terms In Nuclear Science T
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Glossary Of Terms In Nuclear Science T
- 9. Balancing eBooks and Physical Books Glossary Of Terms In Nuclear Science T
 - $\circ\,$ Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Glossary Of Terms In Nuclear Science T
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Glossary Of Terms In Nuclear Science T
 - Setting Reading Goals Glossary Of Terms In Nuclear Science T
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Glossary Of Terms In Nuclear Science T
 - Fact-Checking eBook Content of Glossary Of Terms In Nuclear Science T
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements

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