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web 1 name date per periodic trends lab periodic law states that when the elements are arranged in order of increasing atomic number there is a periodic repetition of their chemical and physical properties elements within the same group have the most similar chemical and physical properties although not identical

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web periodic trends answer key vocabulary atomic radius electron affinity electron cloud energy level group ion ionization energy metal nonmetal nucleus period periodic trends picometer valence electron

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web 1 rank the following elements by increasing atomic radius carbon aluminum oxygen potassium oxygen carbon aluminum potassium 2 rank the following elements by increasing electronegativity sulfur oxygen neon aluminum neon aluminum sulfur oxygen 3 why does fluorine have a higher ionization energy than iodine

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web simulation periodic trends answer key background in this investigation you will examine several periodic trends including atomic radius ionization energy and ionic radius you will

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web 1 how does atomic radius change as you move from left to right across a period explain your answer in terms of the number of protons in the nucleus 2 how does atomic radius change as you move from top to bottom in a group explain your answer in terms of the number of electron shells 3 on the representation of the periodic table below

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web periodic trends worksheet directions use your notes to answer the following questions rank the following elements by increasing atomic radius carbon aluminum oxygen potassium oxygen carbon aluminum potassium rank the following elements by increasing electronegativity sulfur oxygen neon aluminum

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web caragabriel preview chemistry test oct 7 11 43 terms misterlominyo preview terms in this set 25 each element has 3 numbers listed under it which value represents the atomic radius the first number below the diagram what are the units for the atomic radius pm picometers 10 negative twelve meters

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web mar 13 2023 across the periodic table sizes of atoms show the following trends with many irregularities size increases down a group the outermost electrons are in successively more extensive orbitals as n increases

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web similarities and differences among the elements give rise to so called periodic trends both across rows and within columns of the periodic table recognizing periodic trends in the physical and chemical properties of the elements is key to understanding the full value of the periodic table concepts periodic table periodic trends

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web question periodic trends straw lab the purpose of this lab is to create a visual representation of trends on the periodic table you will need answer the following questions and write a formal lab report use the information below to help guide you through writing your lab report 1