

Periodic Table of the Elements



11.01

H

Hydrogen

3.04

Li

Lithium

6.94

Be

Beryllium

9.01

B

Boron

12.01

C

Carbon

14.01

N

Nitrogen

16.00

O

Oxygen

19.00

F

Fluorine

20.18

Ne

Neon

22.99

Na

Sodium

24.31

Mg

Magnesium

26.98

Al

Aluminum

28.09

Si

Silicon

30.97

P

Phosphorus

32.07

S

Sulfur

35.45

Cl

Chlorine

39.96

Ar

Argon

39.10

K

Potassium

40.08

Ca

Calcium

44.96

Sc

Scandium

47.88

Ti

Titanium

50.94

V

Vanadium

52.00

Cr

Chromium

54.94

Mn

Manganese

55.85

Fe

Iron

58.93

Co

Cobalt

58.93

Ni

Nickel

63.55

Cu

Copper

65.39

Zn

Zinc

69.72

Ga

Gallium

72.64

Ge

Germanium

74.92

As

Arsenic

78.97

Se

Selenium

79.90

Br

Bromine

83.80

Kr

Krypton

85.47

Rb

Rubidium

87.62

Sr

Strontium

88.91

Y

Yttrium

91.22

Zr

Zirconium

92.91

Nb

Niobium

95.94

Mo

Molybdenum

98.91

Tc

Technetium

101.07

Ru

Ruthenium

102.91

Rh

Rhodium

106.42

Pd

Palladium

107.87

Ag

Silver

112.41

Cd

Cadmium

114.82

In

Indium

118.71

Sn

Tin

121.76

Sb

Antimony

127.60

Te

Tellurium

126.91

I

Iodine

131.29

Xe

Xenon

132.91

Cs

Cesium

137.33

Ba

Barium

138.91

La

Lanthanum

175.07

Hf

Hafnium

180.95

Ta

Tantalum

183.84

W

Tungsten

186.21

Re

Rhenium

188.91

Os

Osmium

190.23

Ir

Iridium

195.08

Pt

Platinum

196.97

Au

Gold

200.59

Hg

Mercury

204.38

Tl

Thallium

207.2

Pb

Lead

208.98

Bi

Bismuth

209

Po

Polonium

210

At

Astatine

210

Rn

Radon

223

Fr

Francium

226

Ra

Radium

227

Ac

Actinium

232

Th

Thorium

231

Pa

Protactinium

238

U

Uranium

237

Np

Neptunium

244

Pu

Plutonium

243

Am

Americium

247

Cm

Curium

250

Bk

Berkelium

259

Cf

Californium

262

Es

Einsteinium

267

Fm

Fermium

271

Md

Mendelevium

285

No

Nobelium

289

Lr

Lawrencium

atomic number

atomic weight

symbol

black solid

blue liquid

red gas

white synthetically prepared

near ambient temp

alkali metals

alkaline earth metals

transitional metals

other metals

nonmetals

noble gases

LABS

5

6

7

8

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11

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14

15

16

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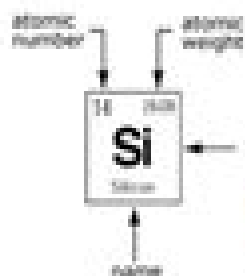
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Legend for element states:

- Black: solid
- Blue: liquid
- Red: gas
- White: synthetically prepared
- Grey: other metals

- alkali metals
- alkaline earth metals
- transitional metals
- other metals
- nonmetals
- noble gases

58 Ce Cerium	59 Pr Praseodymium	60 Nd Neodymium	61 Pm Promethium	62 Sm Samarium	63 Eu Europium	64 Gd Gadolinium	65 Tb Terbium	66 Dy Dysprosium	67 Ho Holmium	68 Er Erbium	69 Tm Thulium	70 Yb Ytterbium	71 Lu Lutetium
90 Th Thorium	91 Pa Protactinium	92 U Uranium	93 Np Neptunium	94 Pu Plutonium	95 Am Americium	96 Cm Curium	97 Bk Berkelium	98 Cf Californium	99 Es Einsteinium	100 Fm Fermium	101 Md Mendelevium	102 No Nobelium	103 Lr Lawrencium

Elements Physical Science

Silas Ellsworth Coleman



Elements Physical Science:

The World of Elements and Their Properties Lynn Van Gorp, 2007-08-03 From compounds to chemical reactions readers will learn all about elements their properties and how they react with other elements in this stunning book that features colorful images and intriguing facts Ionic bonds chemical bonds the Periodic Table of Elements mixtures and solutions are some of the topics that are discussed The accessible glossary and index gives readers the tools they need to better understand the content while a fascinating hands on lab activity will leave readers engaged and excited to learn more

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Addison-Wesley Physical Science Gordon P. Johnson, Bonnie B. Barr, Michael B. Leyden, Addison-Wesley Publishing Company, 1988 Elements of Early Modern Physics J. L. Heilbron, 2024-03-29 *Elements of Early Modern Physics* comprises the two long introductory chapters of J L Heilbron's monumental work *Electricity in the 17th and 18th Centuries A Study of Early Modern Physics* plus a concluding summary of the remaining chapters Heilbron opens with a presentation of the general principles of physical theory and a description of the institutional frameworks in which physics were cultivated in the seventeenth and eighteenth centuries He argues that the single most important contributor to physics in the seventeenth century was the Catholic Church In the first half of the eighteenth century Cartesian and Newtonian physicists disagreed over principles but thought in similar terms and cultivated the same sort of qualitative natural philosophy Work towards an

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