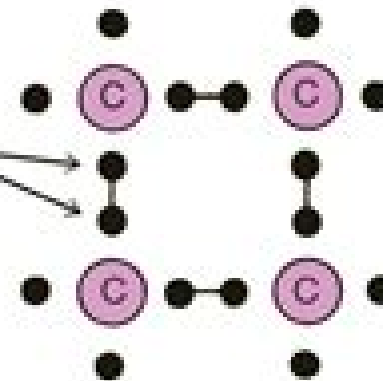
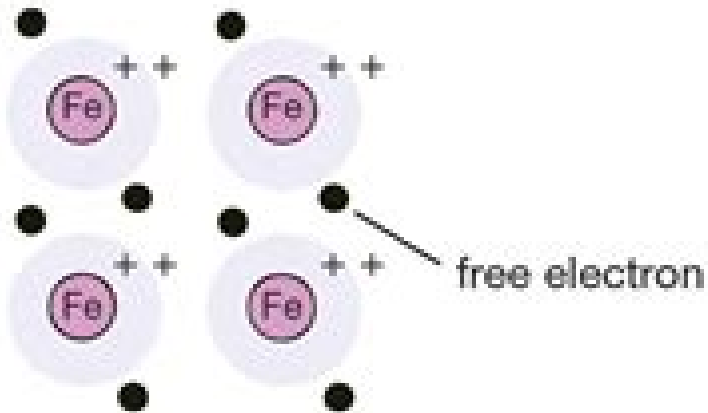


ionic bonding
electron transferred from Na to Cl

shared electrons

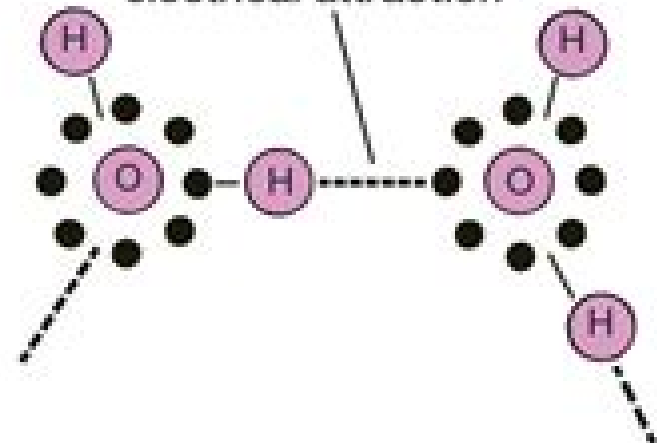


covalent bonding
atoms share electrons



metallic bonding
ions surrounded by free electrons

electrical attraction



molecular bonding
weak electrical attraction binds molecules

Electronic Structure And Chemical Bonding

Zvonimir B. Maksic



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Electronic Structure and the Properties of Solids Walter A. Harrison, 2012-03-08 This text offers basic understanding of the electronic structure of covalent and ionic solids, simple metals, transition metals and their compounds; also explains how to calculate dielectric, conducting, bonding properties.

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Unified Valence Bond Theory of Electronic Structure N. D. Epitotis, 2012-12-06

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Perspectives in Electronic Structure Theory Roman F. Nalewajski, 2012-03-26 The understanding in science implies insights from several different points of view Alternative modern outlooks on electronic structure of atoms and molecules all rooted in quantum mechanics are presented in a single text Together these complementary perspectives provide a deeper understanding of the localization of electrons and bonds the origins of chemical interaction and reactivity behavior the interaction between the geometric and electronic structure of molecules etc In the opening two parts the basic principles and techniques of the contemporary computational and conceptual quantum chemistry are presented within both the wave function and electron density theories This background material is followed by a discussion of chemical concepts including stages of the bond formation processes chemical valence and bond multiplicity indices the hardness softness descriptors of molecules and reactants and general chemical reactivity stability principles The insights from Information Theory the basic elements of which are briefly introduced including the entropic origins and Orbital Communication Theory of the chemical bond are the subject of Part IV The importance of the non additive interference information tools in exploring patterns of chemical bonds and their covalent and ionic components will be emphasized

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Zvonimir B. Maksic, 2012-12-06 The renowned theoretical physicist Victor F Weisskopf rightly pointed out that a real understanding of natural phenomena implies a clear distinction between the essential and the peripheral Only when we reach such an understanding that is to say when we are able to separate the relevant from the irrelevant will the phenomena no longer appear complex but intellectually transparent This statement which is generally valid reflects the very essence of modelling in the quantum theory of matter on the molecular level in particular Indeed without theoretical models one would be swamped by too many details embodied in intricate accurate molecular wavefunctions Further physically justified simplifications enable studies of the otherwise intractable systems and or phenomena Finally a lack of appropriate models would leave myriads of raw experimental data totally unrelated and incomprehensible The present series of books dwells on the most important models of chemical bonding and on the variety of its manifestations In this volume the electronic structure and properties of molecules are considered in depth Particular attention is focused on the nature of intramolecular interactions which in turn are revealed by various types of molecular spectroscopy Emphasis is put on the conceptual and interpretive aspects of the theory in line with the general philosophy adopted in the series

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