

Periodic Chart of the Elements

IA

NOBLE
GASES

THE SYMBOL. Shown in the middle of each block directly below the name of the element. The colour used indicates the physical state of the element under ordinary conditions: black for solids, green for liquids and blue for gases.

THE ATOMIC WEIGHT. Shown below the symbol for each element is the atomic weight as shown in black. The values are taken from the official IUPAC Atomic Weights Of The Elements (2013). For elements not listed in the Report the mass number of the longest lived isotope is shown in brackets.

THE ATOMIC NUMBER. Shown in red in the upper left hand corner.

ELECTRONIC CONFIGURATION. Shown at the upper right as a group of black numbers. When read downwards they indicate the number of electrons normally found in successive energy levels.

IIIB IVB VB VIB VIIB

1

1	H 1.008
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IIA

3	Li 6.941	4	Be 9.012
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2

11	Na 22.990	12	Mg 24.305
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3

19	K 39.098	20	Ca 40.078	21	Sc 44.956	22	Ti 47.88	23	V 50.942	24	Cr 52.00	25	Mn 54.938	26	Fe 55.845	27	Co 58.933	28	Ni 58.69	29	Cu 63.546	30	Zn 65.38	31	Ga 69.723	32	Ge 72.64	33	As 74.922	34	Se 78.96	35	Br 79.904	36	Kr 83.80
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4

37	Rb 85.468	38	Sr 87.62	39	Y 88.906	40	Zr 91.224	41	Nb 92.906	42	Mo 95.94	43	Tc [98]	44	Ru 101.07	45	Rh 102.91	46	Pd 106.42	47	Ag 107.868	48	Cd 112.411	49	In 114.818	50	Sn 118.710	51	Sb 121.757	52	Te 127.6	53	I 126.905	54	Xe 131.29
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5

55	Cs 132.905	56	Ba 137.327	57	La 138.905	72	Hf 178.49	73	Ta 180.948	74	W 183.84	75	Re 186.207	76	Os 190.23	77	Ir 192.222	78	Pt 195.084	79	Au 196.967	80	Hg 200.59	81	Tl 204.383	82	Pb 207.2	83	Bi 208.980	84	Po [209]	85	At [210]	86	Rn [222]
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6

87	Fr [223]	88	Ra [226]	89	Ac [227]	104	Unq [261]	105	Unp [262]	106	Unh [263]	107	Uns [264]
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7

VIII

IB IIB

THE SYMBOL. Shown in the middle of each block directly below the name of the element.
The color used indicates the physical state of the element under ordinary conditions: black for solids, green for liquids and blue for gases.

THE ATOMIC WEIGHT. Shown below the symbol for each element - the atomic weight is shown in black. The values are taken from the official Report on Atomic Weights (J. J. Auer, Chem. Soc. 1913-1914). For elements not listed in the Report the mass number of the longest lived isotope is shown in brackets.

THE ATOMIC NUMBER. Shown in red in the upper left hand corner.

ELECTRONIC CONFIGURATION. Shown at the upper right as a group of black numbers. When read downward they indicate the number of electrons normally found in successive energy levels.

58	Ce 140.12	59	Pr 140.908	60	Nd 144.24	61	Pm [145]	62	Sm 150.36	63	Eu 151.964	64	Gd 157.25	65	Tb 158.925	66	Dy 162.50	67	Ho 164.930	68	Er 167.259	69	Tm 168.930	70	Yb 173.054	71	Lu 174.967
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90	Th 232.038	91	Pa 231.036	92	U 238.029	93	Np [237]	94	Pu [244]	95	Am [243]	96	Cm [247]	97	Bk [247]	98	Cf [251]	99	Es [252]	100	Fm [257]	101	Md [258]	102	No [259]	103	Lr [262]
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Elements Of Old English

Samuel Moore, Thomas A. Knott



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