

Elements: Crystal Structures, High-Pressure Behavior with Implications for the Earth's Interior by Liang-guo Li and William A. Bassett. The Clarendon Press, Oxford University Press, New York, 1986. 250p., £54.00 (ISBN 0-19-503681-4).

AS THE TITLE of the book indicates, the authors have compiled a large amount of very useful data on high-pressure properties of elements, oxides, and silicates. While the authors aimed to provide data on studying the Earth's interior, the properties of material described in the book are not only useful to geophysicists but also to chemists, physicists, metallurgists, and ceramicists. The book consists of five chapters; the first is an introduction and the last an effort of the authors to demonstrate the application of the data in studying the Earth. Sandwiched in-between are three chapters with review and discussion of the available data on the properties of the high-pressure and high-temperature material.

Chapter 1 introduces a variety of subjects, including thermodynamics, crystal chemistry, and experimental techniques. In 23 pages the authors summarized important information relevant to the determination and discussion of the data that will follow in the next three chapters. The chapter is, however, too brief, particularly the section on experimental techniques.

Chapter 2 describes the properties of elements. In this and the other chapters, pressure-temperature phase diagrams have been drawn with no indication of the errors in experimental data. If all the experimental data were given, the usefulness of the book would have been greatly enhanced to researchers interested in modeling the P-T behavior of the solids and melts. However, since all references to the original data are given in the text, the book is still useful to research, albeit some additional effort is required. The references of

the data on elements is essentially complete with numerous references to the original experiments.

The third chapter, running for 103 pages, is concerned with the oxides. As for the element in the second chapter, the oxides are arranged according to their structural forms and complexity of composition. The fourth chapter of the book deals with silicates in the same style. The material covered in these two chapters is quite exhaustive and should be a very important source for designing new experiments or in attempting theoretical modeling of thermodynamic relations for solids or for geophysical-geochemical applications. Chapter 4 also contains an excellent review of the various types of work on oxides, but here (as in other chapters) the original experimental data are not listed in tables or plotted in figures.

While the format/index is good, I could not find "pressure" in the subject-index (which is curious because one of the authors is credited with the discovery of this phase.)

In spite of the minor criticisms I have about the book, it should be clearly stated that the book is a very useful reference for my work. There is no other book available which treats the phase equilibrium data at high-pressure and high-temperature so thoroughly and extensively. All the important information available up to 1986 may be found here. Although there has been quite a rapid progress in the science of mineral-phases during the last several years, with this book a researcher will have to look back at references for only those two years to update his/her library files.

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S. K. Narayan

Comprehensive Kinetics in Aqueous Systems—An Approach Towards an Jacques Barille. Translated from the French by N. P. Kozmenko, A. Kozmenko, and K. S. Akopov. Ellis Horwood, Chichester (distributed by J. Wiley & Sons, 1988. 492p., \$145.00) ISBN 0-85199-208-9/91.

THE UNCERTAINTY in applying the equations of aqueous solutions often impedes the quantitative evaluation of aqueous systems. This nebulous problem is explored in this book through a critical discussion of present methods of solving aqueous reactions, both theoretically and experimentally. It uniquely combines the distribution of solute components not only among simple inorganic and organic species but also on particular surfaces and in organisms and macromolecules.

The first half of the book examines the kinetic and equilibrium factors that determine the time-dependent distribution of species, particularly in complexes. The factors include particle sizes, concentrations of fresh, mature and interstitial sediment water, undissolved mineral stability, complexation constants, ratios, redoxpe ratios, and character of natural organic solutes, for examples. Tables of the values of these factors in various environments are given, together

with useful equations concerning their variations with geologic environment, and detailed examples are presented from natural systems to illustrate the influences of the factors on speciation.

The second half of the book describes experimental methods that are sensitive to speciation, including volumetric, spectroscopy, the ammonia, NMR, ESR, ion exchange, adsorption, and solubility.

This volume is well printed, bound, edited, and indexed. It appears at the front a useful table of notation, necessary for the many abbreviations adopted.

Those geochemists interested in solving the true speciation of surface waters will find this book not a new standard for the comprehensive, balanced treatment of the distribution of aqueous components among all pertinent factors. Although hard, it has a high degree of information. The book of *hydrochemistry* is suitable for the advanced graduate student or the veteran geochemist to whom it is sincerely recommended, provided that he has sufficient time available to cover the manuscript page.

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H. F. Hsieh

Elements Oxides And Silicates High Pressure Phases With Implications For The Earths Interior

Tao Wei



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