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2	<div><div>3</div><div>Li</div><div>Lithium</div></div>	<div><div>4</div><div>Be</div><div>Beryllium</div></div>																	<div><div>10</div><div>Ne</div><div>Neon</div></div>
3	<div><div>11</div><div>Na</div><div>Sodium</div></div>	<div><div>12</div><div>Mg</div><div>Magnesium</div></div>																	<div><div>18</div><div>Ar</div><div>Argon</div></div>
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5	<div><div>37</div><div>Rb</div><div>Rubidium</div></div>	<div><div>38</div><div>Sr</div><div>Strontium</div></div>	<div><div>39</div><div>Y</div><div>Yttrium</div></div>	<div><div>40</div><div>Zr</div><div>Zirconium</div></div>	<div><div>41</div><div>Nb</div><div>Niobium</div></div>	<div><div>42</div><div>Mo</div><div>Molybdenum</div></div>	<div><div>43</div><div>Tc</div><div>Technetium</div></div>	<div><div>44</div><div>Ru</div><div>Ruthenium</div></div>	<div><div>45</div><div>Rh</div><div>Rhodium</div></div>	<div><div>46</div><div>Pd</div><div>Palladium</div></div>	<div><div>47</div><div>Ag</div><div>Silver</div></div>	<div><div>48</div><div>Cd</div><div>Cadmium</div></div>	<div><div>49</div><div>In</div><div>Indium</div></div>	<div><div>50</div><div>Sn</div><div>Tin</div></div>	<div><div>51</div><div>Sb</div><div>Antimony</div></div>	<div><div>52</div><div>Te</div><div>Tellurium</div></div>	<div><div>53</div><div>I</div><div>Iodine</div></div>	<div><div>54</div><div>Xe</div><div>Xenon</div></div>	
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7	<div><div>87</div><div>Fr</div><div>Francium</div></div>	<div><div>88</div><div>Ra</div><div>Radium</div></div>	<div><div>89-103</div><div>Actinide Series</div></div>	<div><div>104</div><div>Rf</div><div>Rutherfordium</div></div>	<div><div>105</div><div>Db</div><div>Dubnium</div></div>	<div><div>106</div><div>Sg</div><div>Seaborgium</div></div>	<div><div>107</div><div>Bh</div><div>Bohrium</div></div>	<div><div>108</div><div>Hs</div><div>Hassium</div></div>	<div><div>109</div><div>Mt</div><div>Meitnerium</div></div>	<div><div>110</div><div>Ds</div><div>Darmstadtium</div></div>	<div><div>111</div><div>Rg</div><div>Roentgenium</div></div>	<div><div>112</div><div>Cn</div><div>Copernicium</div></div>	<div><div>113</div><div>Nh</div><div>Nihonium</div></div>	<div><div>114</div><div>Fl</div><div>Flerovium</div></div>	<div><div>115</div><div>Mc</div><div>Moscovium</div></div>	<div><div>116</div><div>Lv</div><div>Livermorium</div></div>	<div><div>117</div><div>Ts</div><div>Tennessine</div></div>	<div><div>118</div><div>Og</div><div>Oganesson</div></div>	
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Atomic Number →

1

1.008

← Average Atomic Mass

Symbol →

H

Name →

Hydrogen

Alkali Metals

Alkaline Earth Metals

Transition Metals

Other Metals

Metalloids

Non-metals

Halogens

Noble Gases

Lanthanides

Actinides

Elements Of Chemical Physics

JOSIAH PARSONS. COOKE



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