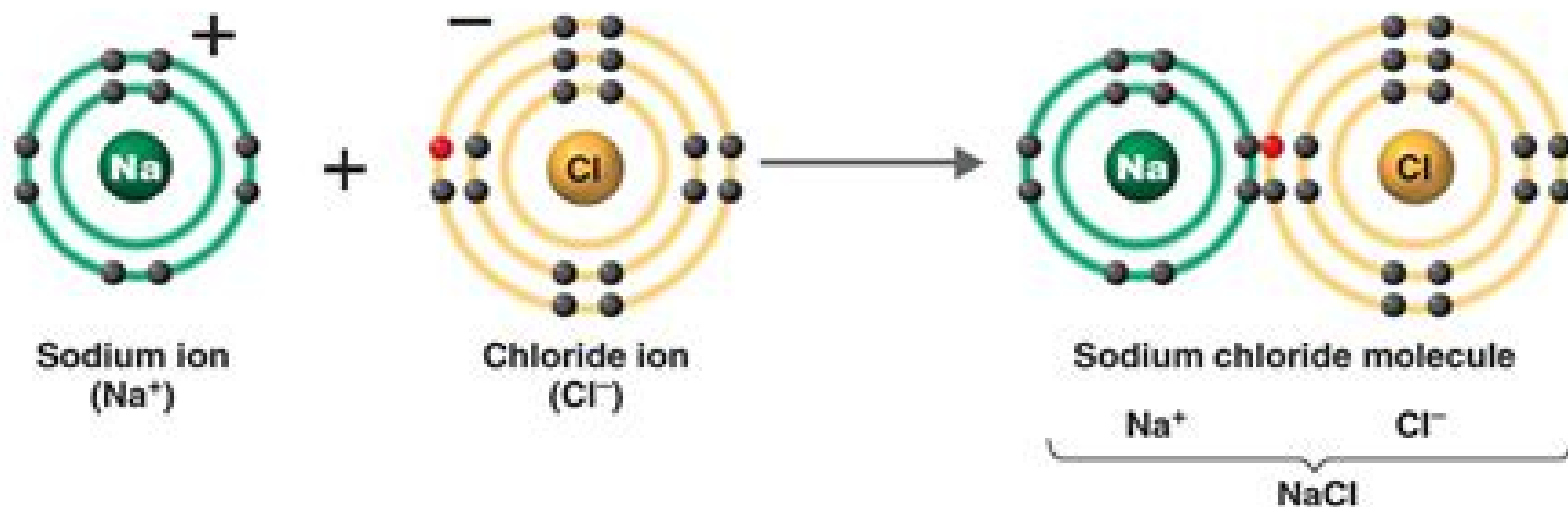


**(a)** A sodium atom (Na) loses one electron to an electron acceptor and forms a sodium ion (Na<sup>+</sup>). A chlorine atom (Cl) accepts one electron from an electron donor to become a chloride ion (Cl<sup>-</sup>).



**(b)** The sodium and chloride ions are attracted because of their opposite charges and are held together by an ionic bond to form a molecule of sodium chloride.

# Electronic Structure Chemical Bonding

**DP Hallahan**



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