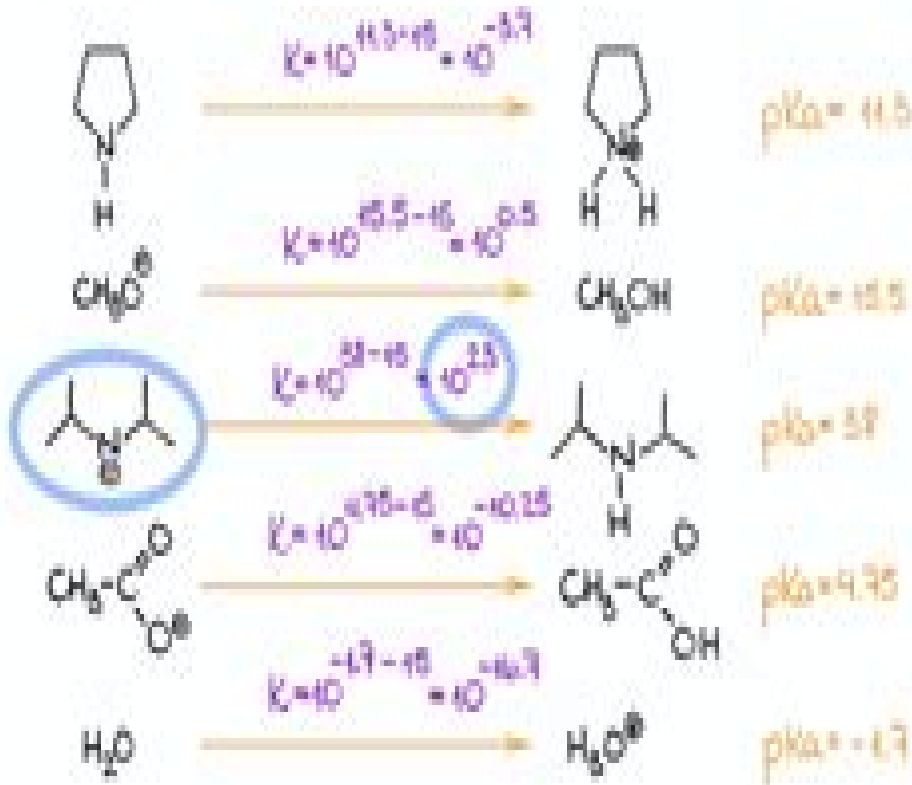


To solve this question, we need to consider the following equilibrium:



1. Set up the stem for your acid-base equilibrium

2. Write the conjugate acids for your bases

3. Find the pKa values of those conjugate acids

4. Calculate the K values by using

$$K = 10^{\text{pKa}(\text{C.B.}) - \text{pKa}(\text{A})}$$

equation

5. Compare your K values to the K value in the question

# Equilibrium Freshman Chemistry Problems And How To Solve Them Part 2

**Michell J. Sienko**



## **Equilibrium Freshman Chemistry Problems And How To Solve Them Part 2:**

## **Equilibrium Freshman Chemistry Problems And How To Solve Them Part 2** Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the ability of words has become much more evident than ever. They have the ability to inspire, provoke, and ignite change. Such may be the essence of the book **Equilibrium Freshman Chemistry Problems And How To Solve Them Part 2**, a literary masterpiece that delves deep into the significance of words and their affect on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book's key themes, examine its writing style, and analyze its overall affect on readers.

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