

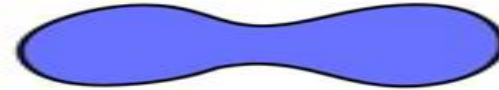
EQUILIBRIUM

a balanced system created by opposing forces

for any set of reactants and products we get 2 different reactions

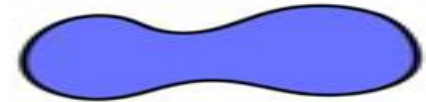
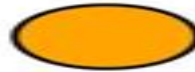


reactants



product

a forward reaction:

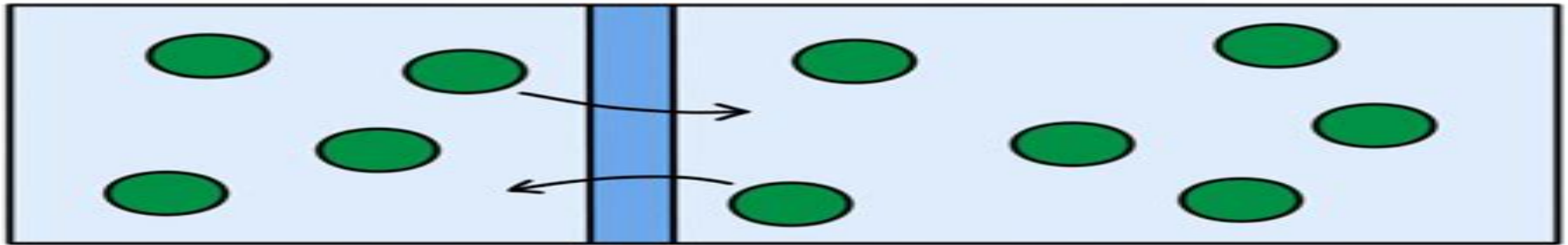


and a reverse reaction:

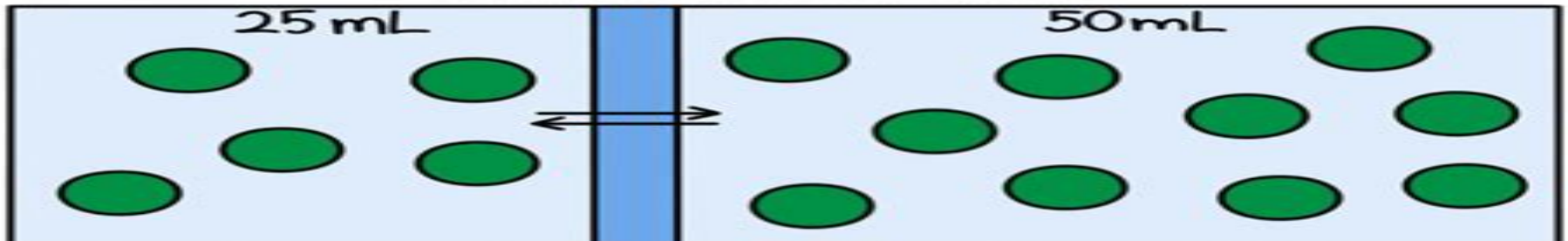


at equilibrium, both reactions are happening at the same rate
there is no net change in the amount of reactant or product

equilibrium can also be seen in terms of transport
across the cell membrane



as some molecules move into the cell, others move out
at equilibrium, there is an equal concentration of solute
inside the cell and outside



make sure to factor in the volume because it's the
concentration that matters, not the number of molecules

Equilibrium Three Volume Set

Munmaya Mishra



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