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Electronic Structure and Chemical Binding Oscar Knefler Rice,1946 *Electronic Structure and Chemical Binding* Oscar K. Rice,1983-07 **Electronic Structure and Chemical Binding** Oscar Knefler Rice,1940 *Electronic Structure and Chemical Bonding* Oscar Knefler Rice,1940 **From Chemical Philosophy to Theoretical Chemistry** Mary Jo Nye,1994-03-01 How did chemistry and physics acquire their separate identities and are they on their way to losing them again Mary Jo Nye has written a graceful account of the historical demarcation of chemistry from physics and subsequent reconvergences of the two from Lavoisier and Dalton in the late eighteenth century to Robinson Ingold and Pauling in the mid twentieth century Using the notion of a disciplinary identity analogous to ethnic or national identity Nye develops a theory of the nature of disciplinary structure and change She discusses the distinctive character of chemical language and theories and the role of national styles and traditions in building a scientific discipline Anyone interested in the history of scientific thought will enjoy pondering with her the question of whether chemists of the mid twentieth century suspected chemical explanation had been reduced to physical laws just as Newtonian mechanical philosophers had envisioned in the eighteenth century Catalog of Books and Reports in the Bureau of Mines Technical Library, Pittsburgh, Pa United States. Bureau of Mines. Technical Library, Pittsburgh,1968 **Agricultural Library Notes** ,1940 *Inorganic Chemistry for Geochemistry and Environmental Sciences* George W. Luther, III,2016-08-01 Inorganic Chemistry for Geochemistry and Environmental Sciences Fundamentals and Applications discusses the structure bonding and reactivity of molecules and solids of environmental interest bringing the reactivity of non metals and metals to inorganic chemists geochemists and environmental chemists from diverse fields Understanding the principles of inorganic chemistry including chemical bonding frontier molecular orbital theory electron transfer processes formation of nano particles transition metal ligand complexes metal catalysis and more are essential to describe earth processes over time scales ranging from 1 nanosec to 1 Gigayr Throughout the book fundamental chemical principles are illustrated with relevant examples from geochemistry environmental and marine chemistry allowing students to better understand environmental and geochemical processes at the molecular level Topics covered include Thermodynamics and kinetics of redox reactions Atomic structure Symmetry Covalent bonding and bonding in solids and nanoparticles Frontier Molecular Orbital Theory Acids and bases Basics of transition metal chemistry including Chemical reactivity of materials of geochemical and environmental interest Supplementary material is provided online including PowerPoint slides problem sets and solutions Inorganic Chemistry for Geochemistry and Environmental Sciences is a rapid assimilation textbook for those studying and working in areas of geochemistry inorganic chemistry and environmental chemistry wishing to enhance their understanding of environmental processes from the molecular level to the global level *Catalog of Copyright Entries. Third Series* Library of Congress. Copyright Office,1972 *Pure and Applied Science Books, 1876-1982* ,1982 Over 220 000 entries representing some 56 000 Library of Congress

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Spice,2016-01-22 *Chemical Binding and Structure* describes the chemical binding and structure in terms of current chemical theory This book is composed of 13 chapters and starts with a presentation of the principles of the old and modified quantum theory and its application The next chapters cover some basic topics related to chemical binding and structure including electrons the periodic table the electrovalent and covalent bonds and molecular geometry These topics are followed by discussions on the nature of the bond in transition metal complexes electronic and crystal structure crystallinity and other states of matter The concluding chapters are devoted to some analytical techniques for structure determination such as diffraction and spectroscopic methods This book is of value to high school and college chemistry teachers and students

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Dronskowski,2023-10-04 Modern DFT simulations of solids and molecules are typically based on the mighty plane wave pseudopotential combination Despite being numerically efficient it does not allow for chemical conclusions unless the electronic structure is unitarily transformed into atomic orbitals This primer for chemists and as well for physicists and engineers shows how to simply extract the chemistry and hence truly understand a plethora of real world materials The goal of this humorous primer entertaining to read is to truly serve but not repel the reader Recent in person and also virtual summer schools in Europe and Asia have demonstrated the need for such a primer also to be used for self training Study of New Ternary Rare-Earth Intermetallic Germanides with Polar Covalent Bonding Riccardo Freccero,2020-11-16 The thesis focuses on the syntheses structural characterizations and chemical bonding analyses for several ternary R M Ge R rare earth metal M another metal intermetallics The challenges in understanding the main interactions governing the chemistry of these compounds which lead to our inability to predict their formation structure and properties are what provided the motivation for this study In particular the R₂MGe₆ M Li Mg Al Cu Zn Pd Ag R₄MGe₁₀ x M Li Mg R₂Pd₃Ge₅ Lu₅Pd₄Ge₈ Lu₃Pd₄Ge₄ and Yb₂PdGe₃ phases were synthesized and structurally characterized Much effort was put into the stabilization of metastable phases employing the innovative metal flux method and into the accurate structure solution of twinned crystals Cutting edge position space chemical bonding techniques were combined with new methodologies conceived to correctly describe the Ge M Ge La and also La M polar covalent interactions for the La₂MGe₆ M Li Mg Al Cu Zn Pd Ag series The present results constitute a step forward in our comprehension of ternary germanide chemistry as well as providing a good playground for further investigations **Catalogue of the University of Michigan** University of Michigan,1949

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