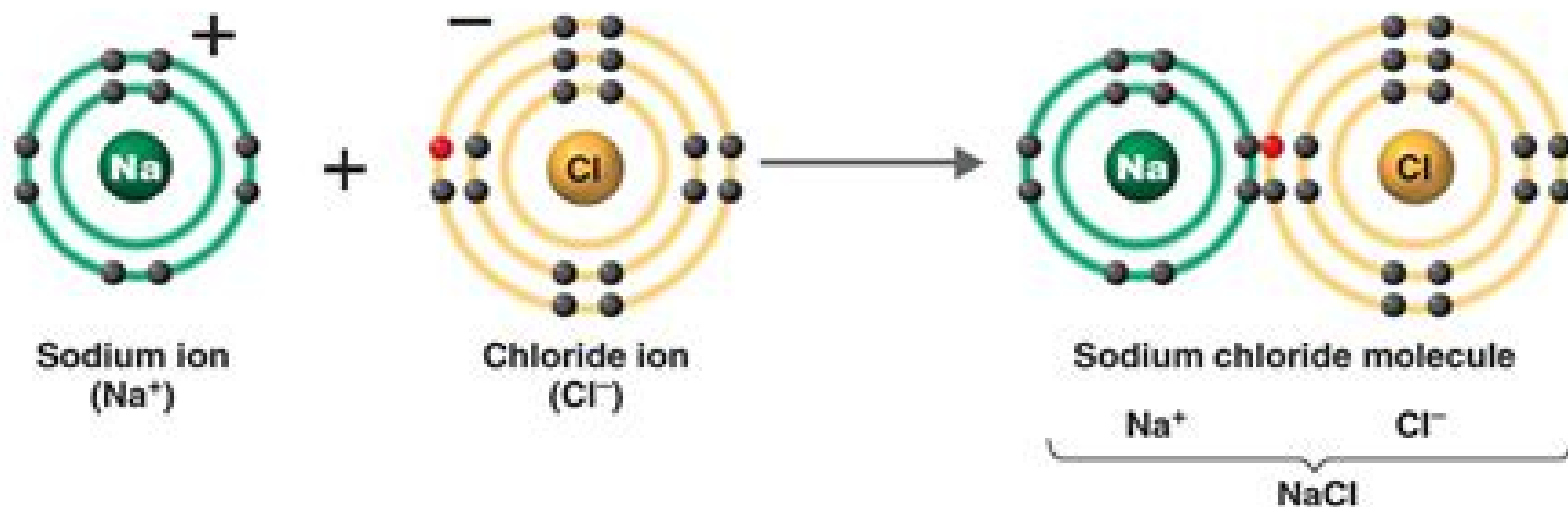


(a) A sodium atom (Na) loses one electron to an electron acceptor and forms a sodium ion (Na⁺). A chlorine atom (Cl) accepts one electron from an electron donor to become a chloride ion (Cl⁻).



(b) The sodium and chloride ions are attracted because of their opposite charges and are held together by an ionic bond to form a molecule of sodium chloride.

Electronic Structure Chemical Bonding

Gregory S. Rohrer

A decorative red circular graphic with a gradient, appearing as a partial circle or a stylized 'C' shape, located to the right of the author's name.

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